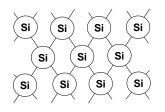
Start with a silicon substrate.

Silicon has 4 valence electrons, and therefore a "lattice" structure:



Each atom bonds with 4 neighbors.

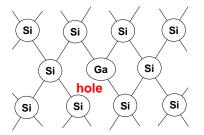
No free electrons (poor conductor) unless you heat it up (semi conductor).

Other Group IV elements can be used for substrate, but they are harder to come by (carbon lattice for instance...)

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DOPING

Make silicon a better conductor by adding Group III or Group V elements: process called **doping**



Add Group V elements like arsenic or phosphorus to get **n-type material**

Note that the As atom has 5 valence electrons; it has an unbonded electron

Si Si Si Si Si e As Si Si Si Si Si

Add Group III elements like gallium or indium to get **p-type material**

Note that the Ga atom has only 3

missing bond is called a hole

valence electrons with which to bond;

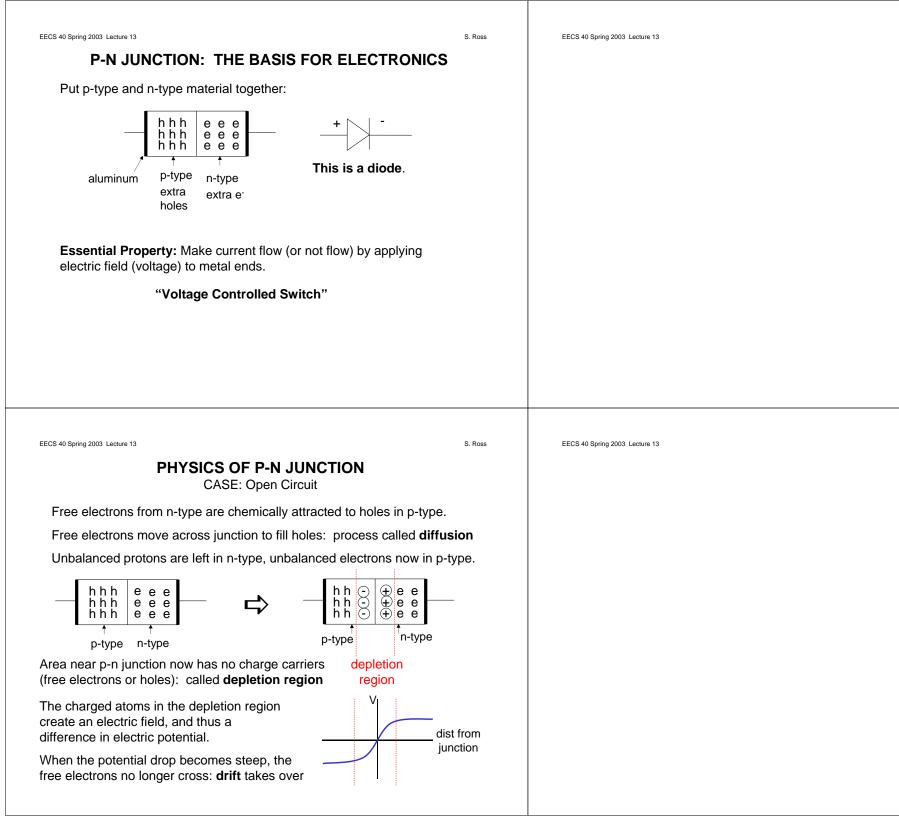
Materials are electrically neutral! Equal number of protons and electrons!

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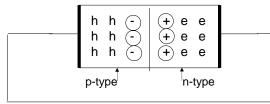
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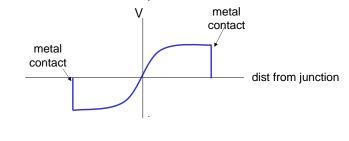


CASE: Short Circuit

Is there a current when I short a diode? Is KVL violated? NO.

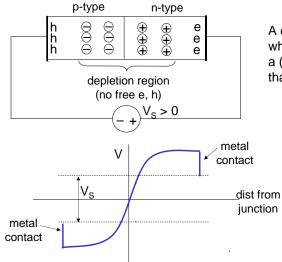


At metal-semiconductor junction, potential changes to balance device. Electrons in metal can redistribute easily to do this.



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PHYSICS OF P-N JUNCTION CASE 2: Reverse Bias



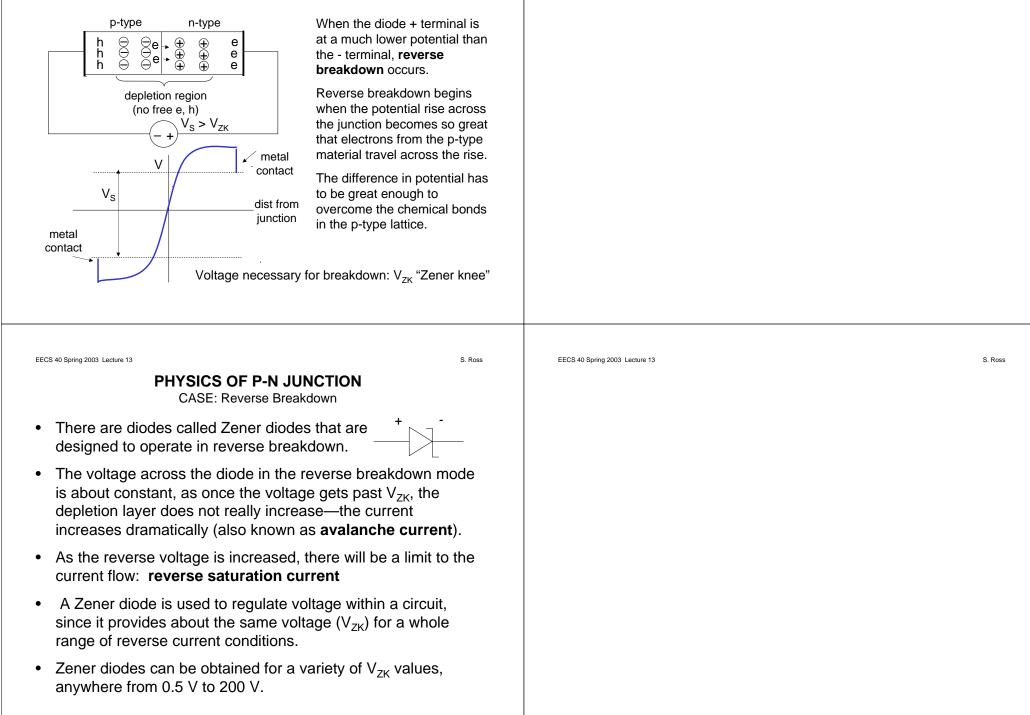
A diode is in **reverse bias** mode when the + (p-type) terminal is at a (moderately) lower potential than the - (n-type) terminal.

> Electrons bunch up by positive metal contact, but few cross through wire because of potential drop between contacts.

A tiny "leakage current" flows due to these few stray electrons, but basically zero current flow.

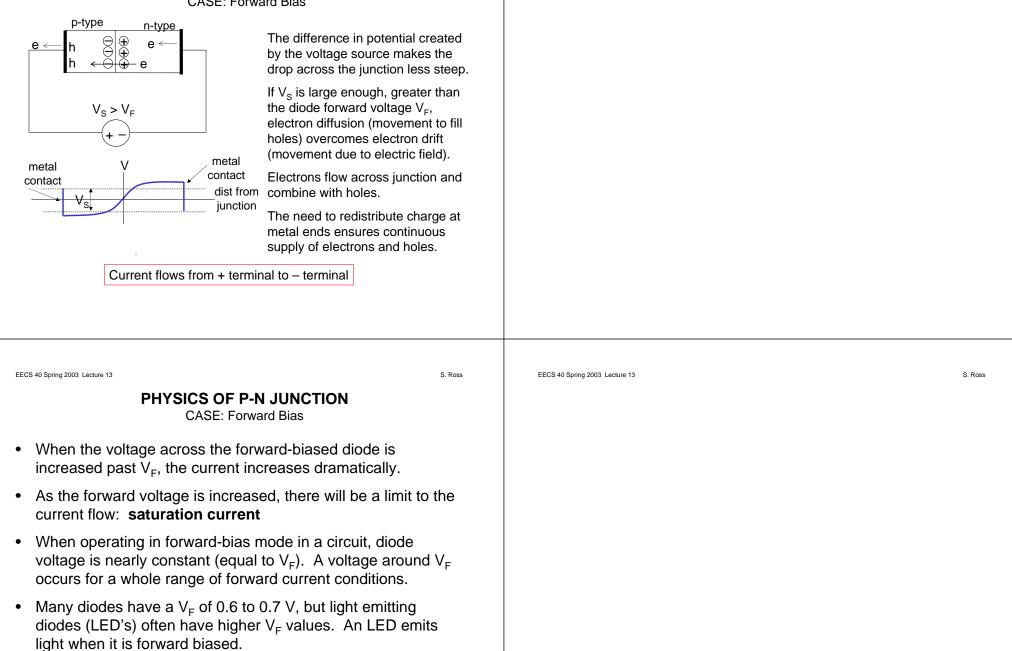


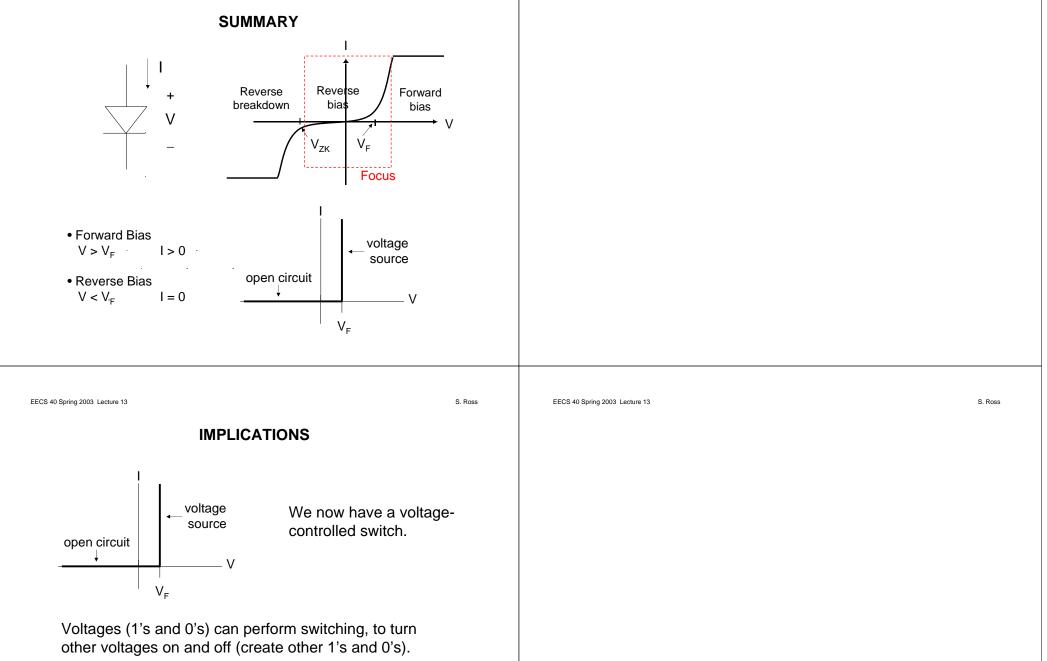
CASE: Reverse Breakdown



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This is the basis of electronic computation.