



EE 105 Microelectronic Devices & Circuits Fall 2018

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Lecture 10 Supplement



Periodic Table & Silicon Semiconductor Elements

Handwritten annotations on the periodic table:

- Bohr Model:** A diagram of Silicon (Si) showing a nucleus with 14 protons and 14 electrons in four shells. The outermost shell is labeled "not set -> so unhappy!".
- Classification:** A green arrow labeled "metal" points to the left side of the table.
- Notes:** "depends" is written in red above elements B, C, N, O, P, and As.
- Circled Elements:**
 - Red circles around Na, Al, Si, and As.
 - Green circles around Ti, Cu, and W.
 - Blue circles around Al and Si.

1 H																	2 He	
3 Li	4 Be																	10 Ne
11 Na	12 Mg																	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
55 Cs	56 Ba	57 La	* 72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
87 Fr	88 Ra	89 Ac	* 104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og	
		* 58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu			
		* 90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr			