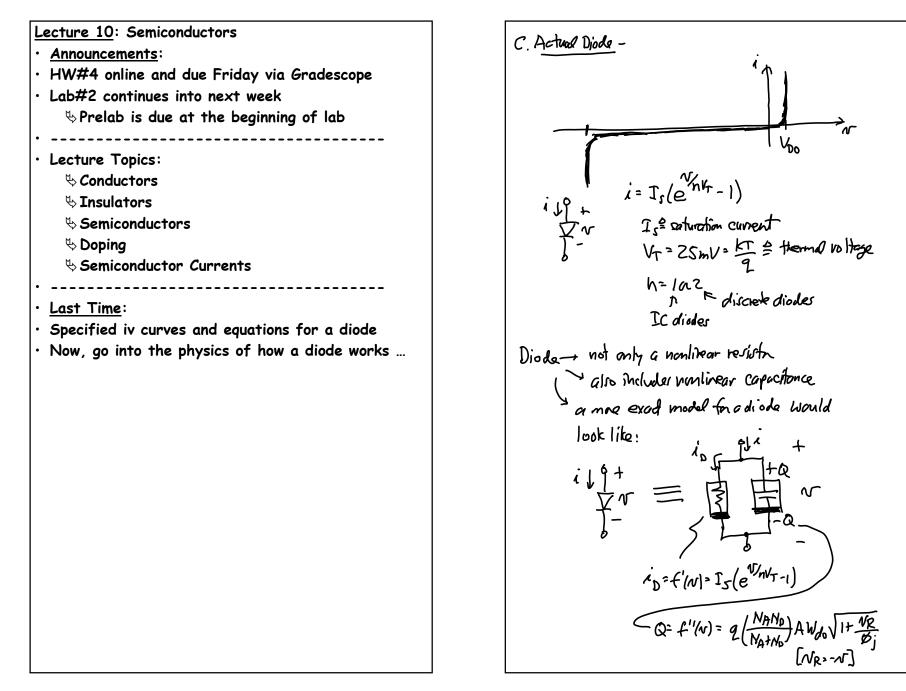
### <u>EE 105</u>: Microelectronic Devices & Circuits <u>Lecture 10w</u>: Semiconductors



### EE 105: Microelectronic Devices & Circuits Lecture 10w: Semiconductors

#### Semiconductors:

- To better understand the physical operation of diodes (and later, transistors), need to understand semiconductors
- Best to describe them in the context of other materials, like conductors and insulators

#### Materials:

- Made up of atoms
- ♦ In solids, the atoms often bond together in a regular lattice
- ♦ The atoms in the lattice happiest in a lowest energy state, i.e., with filled orbitals
- ♦ Go to periodic table supplement

### 1. Conductors:

- Close-packed atoms in a cloud of electrons
- Ex. Na (sodium) a metal



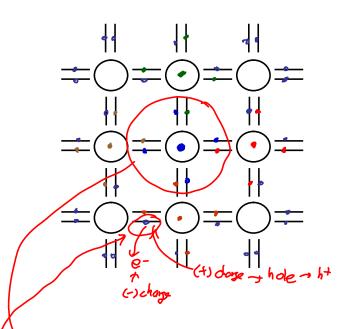
& Orbital below valence shell already filled, so ecan leave and be shared by all atoms in the solid

Va

hucleus (11 protes)

# 2. Insulators:

- Held together very strongly in a regular lattice by strong covalent bonds
- · Lowest energy state when the valence shells of each atom are filled



- Atom happy w/ an effectively filled valence band, so e's not free to move about
- An increase in energy can allow an  $e^-$  to break free into a higher energy free state in which an electric field can move the e<sup>-</sup>
- For example, high enough temperature generates a free  $e^-$  and  $h^+$  pair, each of which is free to move under an applied electric field

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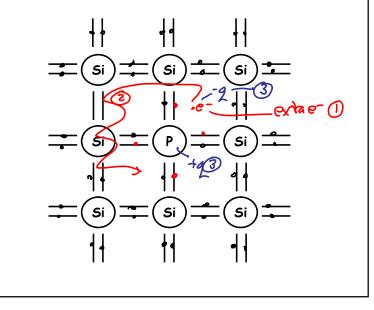
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#### 3. Semiconductors:

- Basically the same as insulators, except they require smaller temperatures to free e<sup>-</sup>s
- For most purposes, they are just like insulators ... until they are doped, at which point they become like metals

#### Doping:

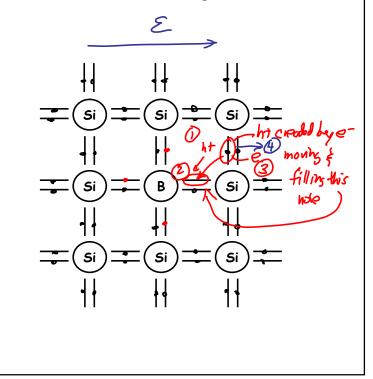
- A semiconductor converts to a conductor when one adds certain impurities that substitute for Si atoms
- Type types of substitutional impurities
  - bonors
  - $\clubsuit$  Acceptors
- 1. <u>Donors</u>:
- Elements with 5 valence e<sup>-'</sup>s, e.g., phosphorous (P) and Arsenic (As)



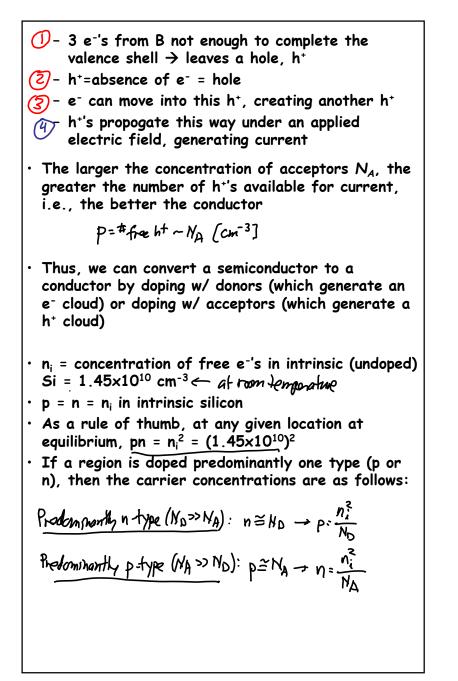
- O- 4 e<sup>-'s</sup> from P contribute to covalent bonds, leaving one extra e<sup>-</sup>
- 🜔 extra e<sup>-</sup> can now move around
- 3)- When e<sup>-</sup> moves away from the donor atom (in this case, P), the donor atom effectively represents a (+) static charge
- The larger the concentration of donors  $N_D$ , the greater the number of e<sup>-'s</sup> available to generate the e<sup>-</sup> cloud, i.e., the better the conductor

n = 10f free e-4 ~ N<sub>0</sub> [cm<sup>-3</sup>]

- 2. Acceptors:
- Elements w/ 3 valence e<sup>-'</sup>s, e.g., Boron (B)



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