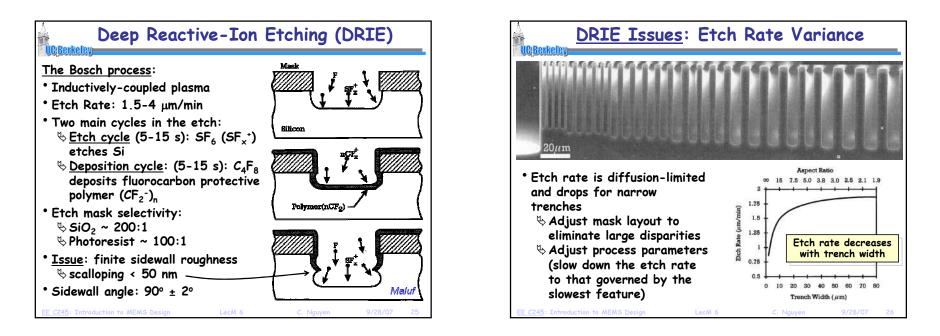
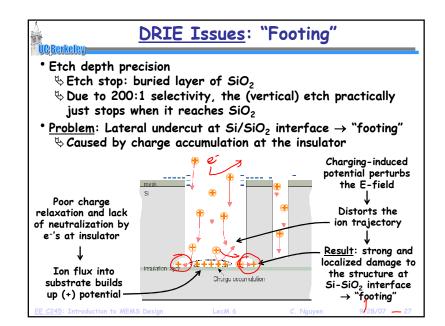
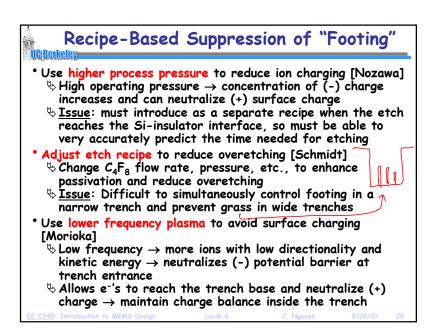
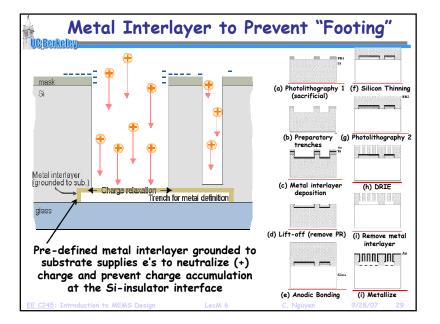
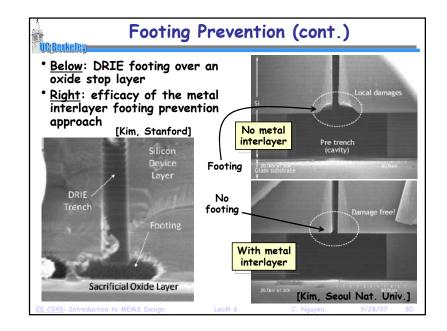
CTN 2/18/16

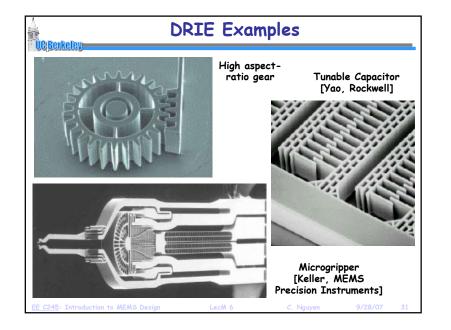


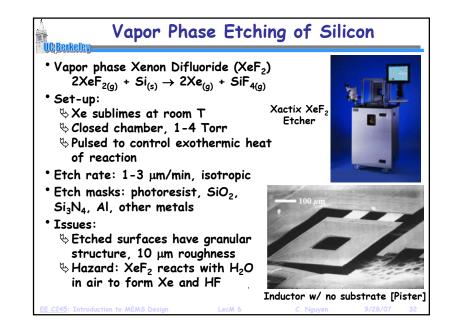




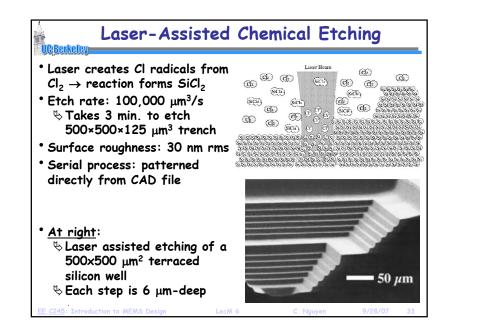


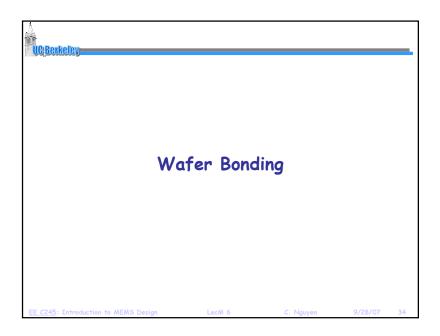


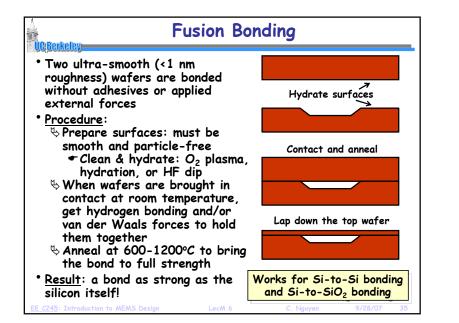


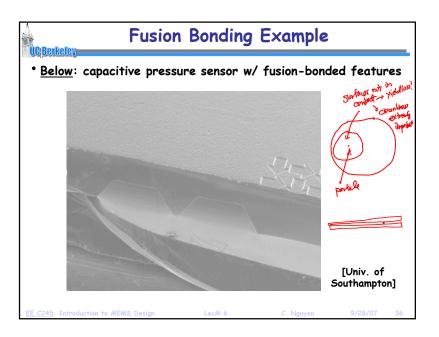






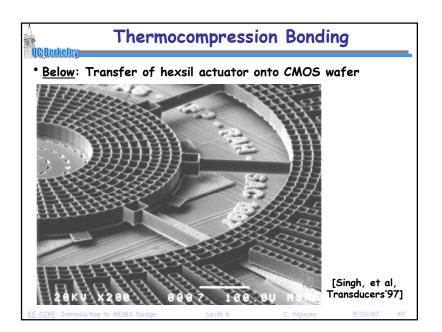


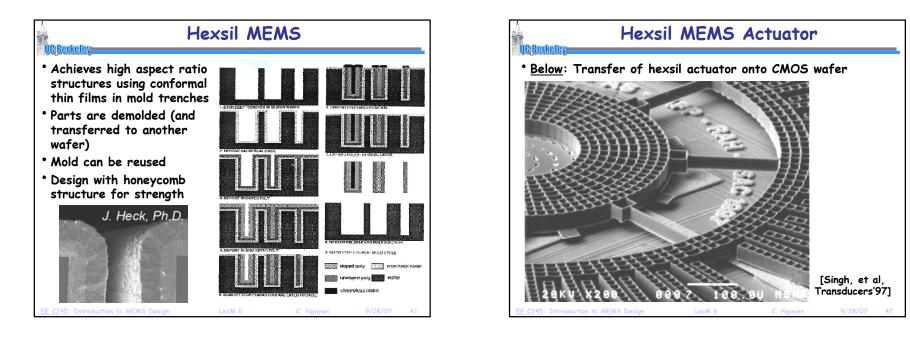


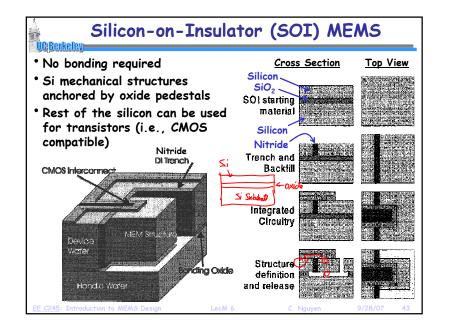


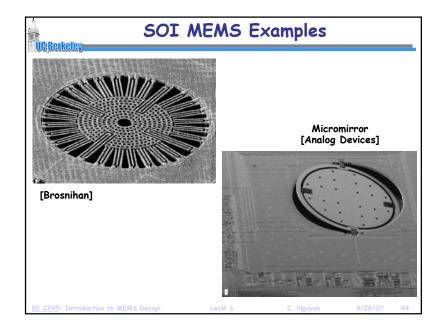
Anodic Bonding • Bonds an electron conducting material (e.g., Si) to an ion conducting material (e.g., sodium glass = Pyrex) Glass Silleon Procedure/Mechanism: Het plate ♦ Press Si and glass together Selevate temperature: 180-500°C ♦ Apply (+) voltage to Si: 200-Temperature 1500V (+) voltage repels Na⁺ ions from the glass surface Pressu • Get net (-) charge at glass retain yield surface Voltage Attractive force between (+) Si and (-) glass \rightarrow intimate. contact allows fusing at elevated temp. Surrent drops to zero when bonding is complete

Anodic Bonding (cont.) Personal • Advantage: high pressure of electrostatic attraction smoothes out defects • Below: 100 mm wafers, Pyrex glass 500 μm-thick, 430°C, 800V, N₂ @ 1000 mbar • Only center bond pin active • Only center bond pin









The SCREAM Process UGB Pootoresist • <u>SCREAM</u>: Single Crystal Reactive Etching and Metallization process xida Silicon substrate 1. Deposit cride and photoresist 2. Lithography and oxide etch 3. Silicon etch 4. Coat sidewalls with PECVD oxide Suspended beam 6. Plasma etch in SF. 5. Remove exide at bottom and etch silicos to release structures