

EE C245 - ME C218 Introduction to MEMS Design Fall 2009

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Lecture Module 6: Bulk Micromachining

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LecM 6

C. Nguyei

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Lecture Outline

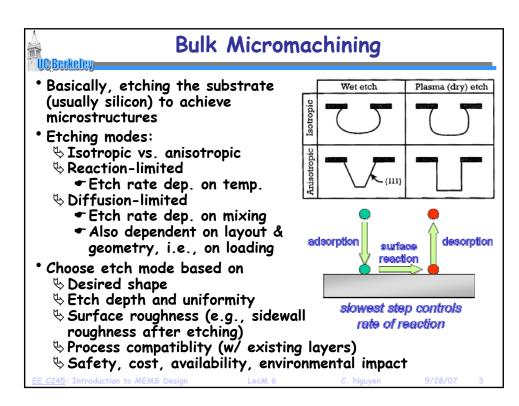
- Reading: Senturia Chpt. 3, Jaeger Chpt. 11, Handouts:
 "Bulk Micromachining of Silicon"
- Lecture Topics:
 - Sulk Micromachining
 - Shanisotropic Etching of Silicon
 - **⇔** Boron-Doped Etch Stop
 - **♥ Electrochemical Etch Stop**
 - **♦ Isotropic Etching of Silicon**
 - ♦ Deep Reactive Ion Etching (DRIE)

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Mechanical Properties of Silicon

- Crystalline silicon is a hard and brittle material that deforms elastically until it reaches its yield strength, at which point it breaks.
 - Tensile yield strength = 7 GPa (~1500 lb suspended from 1 mm²)
 - ♦ Young's Modulus near that of stainless steel

 - ♦ Mechanical properties uniform, no intrinsic stress
 - ♦ Mechanical integrity up to 500°C
 - \$ Good thermal conductor
 - ♦ Low thermal expansion coefficient
 - ⇔ High piezoresistivity

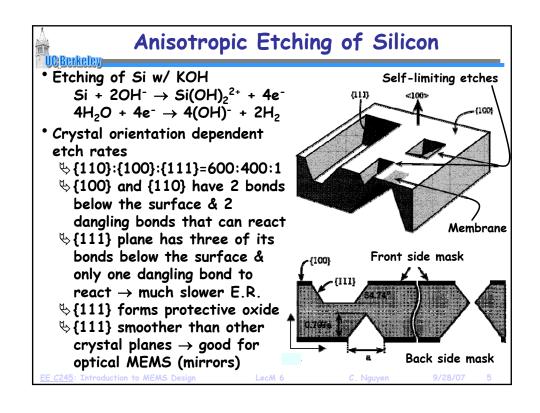
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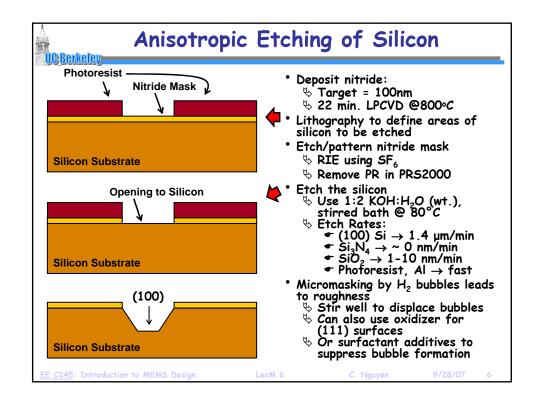
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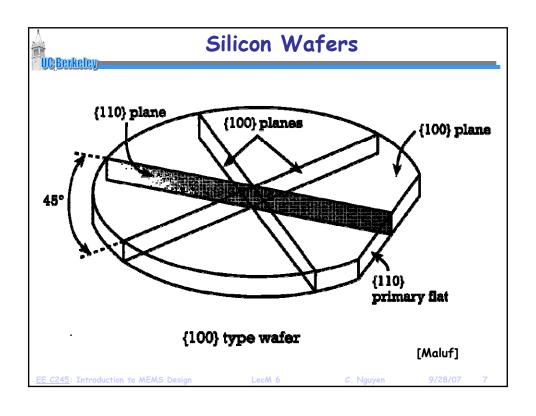
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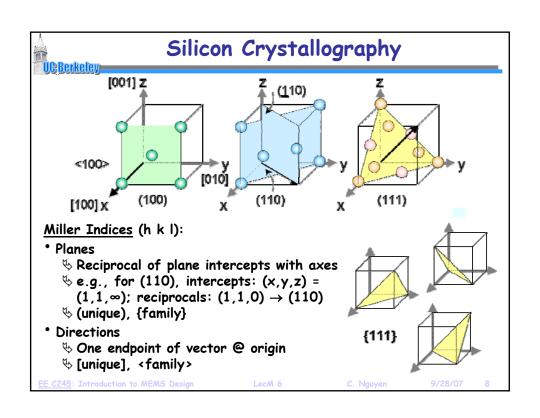
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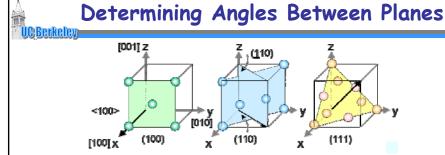
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* The angle between vectors [abc] and [xyz] is given by:

$$ax + by + cz = |(a,b,c)| \cdot |(x,y,z)| \cdot \cos \theta$$

$$\theta_{(a,b,c),(x,y,z)} = \cos^{-1} \left[\frac{ax + by + cz}{|(a,b,c)| \cdot |(x,y,z)|} \right]$$

- $^{\bullet}$ For {100} and {110} \rightarrow 45°
- For $\{100\}$ and $\{111\} \rightarrow 54.74^{\circ}$
- For {110} and {111} \rightarrow 35.26°, 90°, and 144.74°

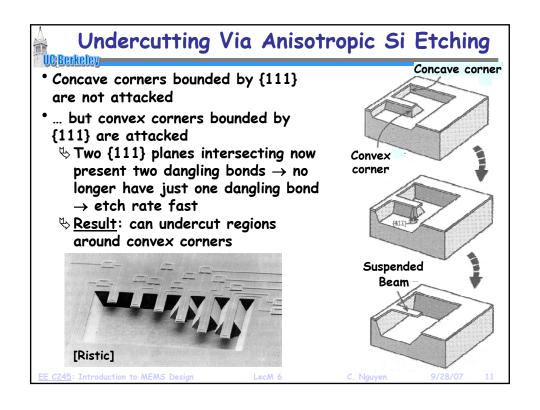
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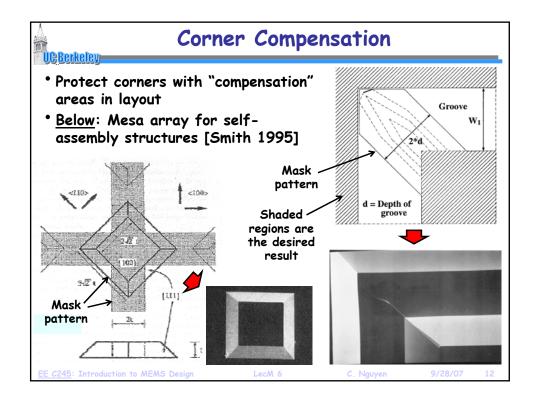
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Silicon Crystal Origami UC Berkeley {110} {100} {110} (100) {100} {110} {110} {110} Silicon fold-up cube • Adapted from Profs. Kris 110 eno {100} {100} {100} Pister and Jack Judy Print onto transparency (111) {110} {110} {110} Assemble inside out Visualize crystal plane {100} orientations, intersections, and directions [Judy, UCLA]





Other Anisotropic Silicon Etchants

- TMAH, Tetramethyl ammonium hydroxide, 10-40 wt.% (90°C)

 - \$ Al safe, IC compatible
 - ♦ Etch ratio (100)/(111) = 10-35
 - $\$ Etch masks: SiO_2 , $Si3N_4 \sim 0.05$ -0.25 nm/min
 - ♦ Boron doped etch stop, up to 40× slower
- EDP (115°C)
 - \$ Carcinogenic, corrosive

 - ♦ Al may be etched
 - ♥ R(100) > R(110) > R(111)
 - \$\infty\$Etch ratio (100)/(111) = 35
 - ♦ Etch masks: SiO₂ ~ 0.2 nm/min, Si₃N₄ ~ 0.1 nm/min
 - ♦ Boron doped etch stop, 50× slower

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13

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Boron-Doped Etch Stop

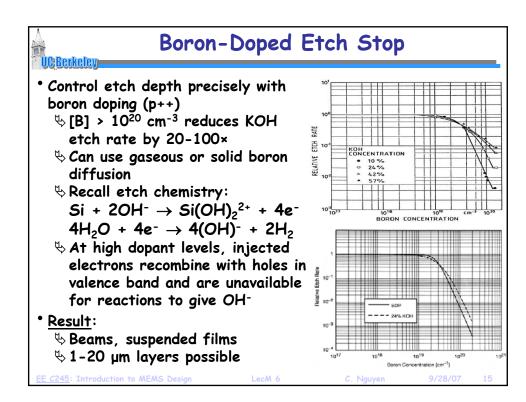
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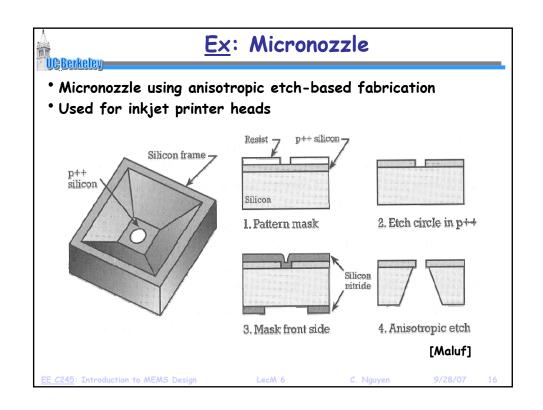
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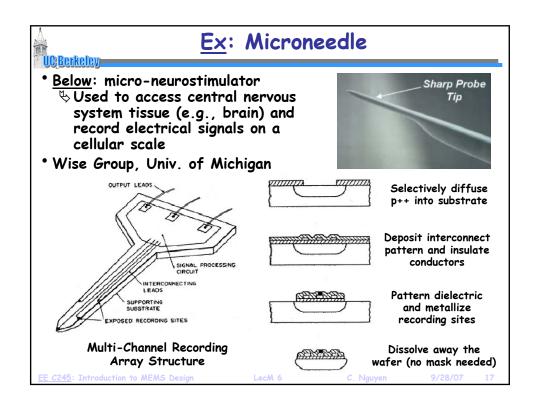
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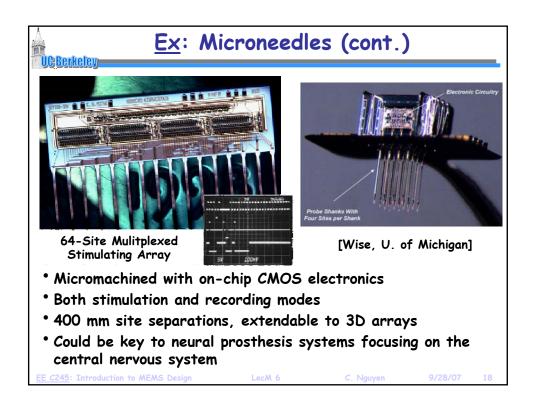
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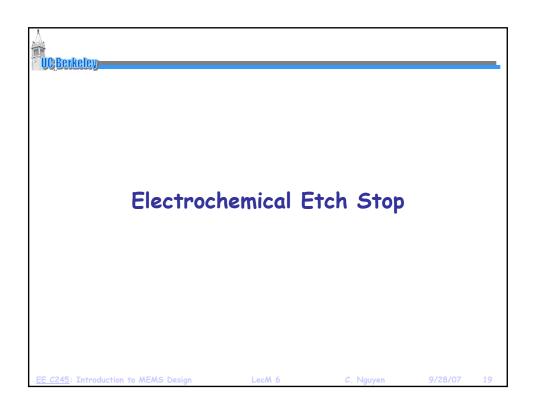
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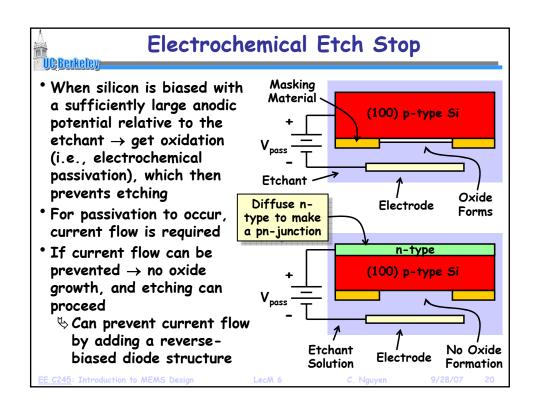


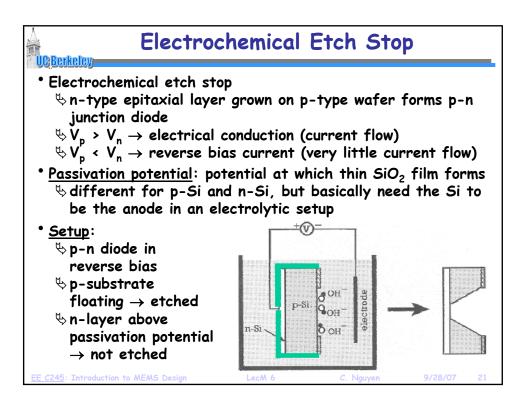


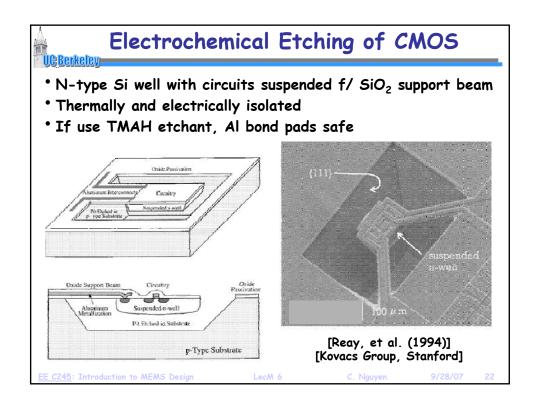


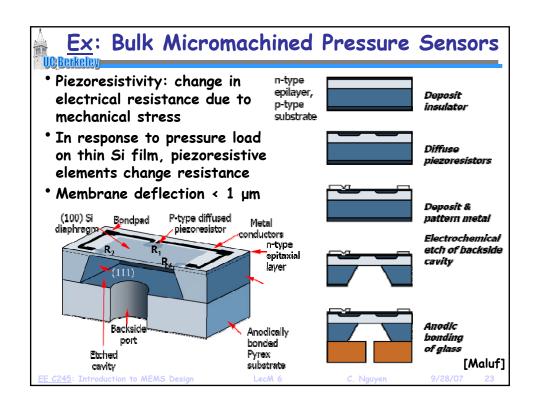


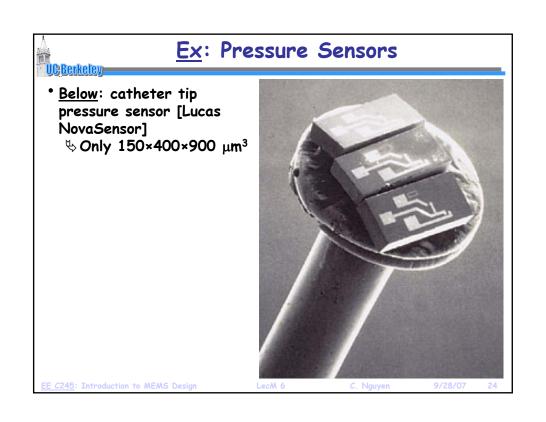


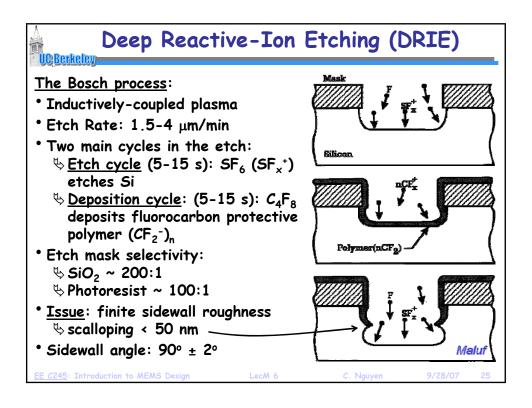


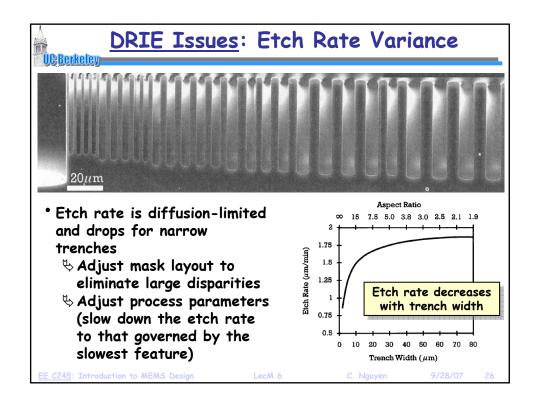


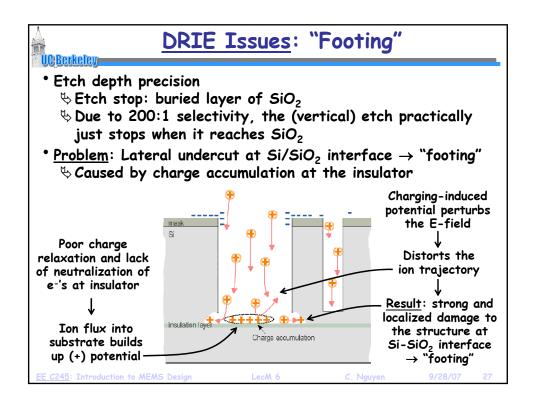












Recipe-Based Suppression of "Footing"

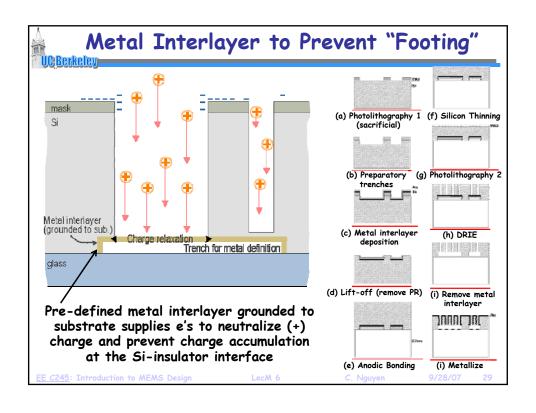
- - ⇒ riigh operating pressure → concentration of (-) ions increases and can neutralize (+) surface charge
 - ➡ <u>Issue</u>: must introduce as a separate recipe when the etch reaches the Si-insulator interface, so must be able to very accurately predict the time needed for etching
- Adjust etch recipe to reduce overetching [Schmidt]
 - $\$ Change C_4F_8 flow rate, pressure, etc., to enhance passivation and reduce overetching
 - ➡ <u>Issue</u>: Difficult to simultaneously control footing in a narrow trench and prevent grass in wide trenches
- Use lower frequency plasma to avoid surface charging [Morioka]
 - ♦ Low frequency → more ions with low directionality and kinetic energy → neutralizes (-) potential barrier at trench entrance
 - ♦ Allows e^{-'}s to reach the trench base and neutralize (+) charge → maintain charge balance inside the trench

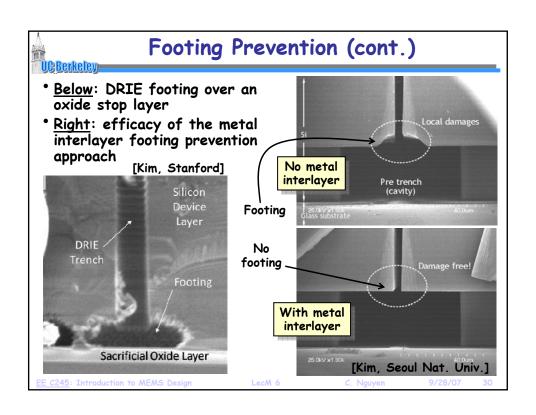
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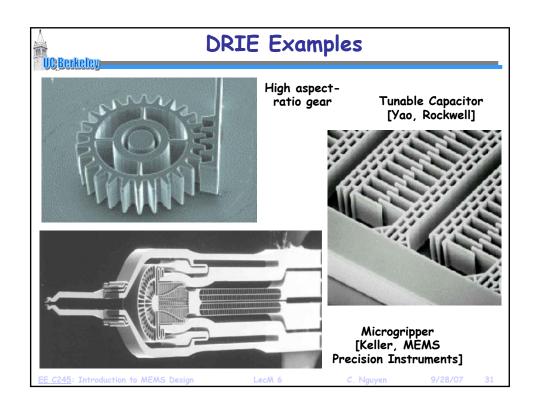
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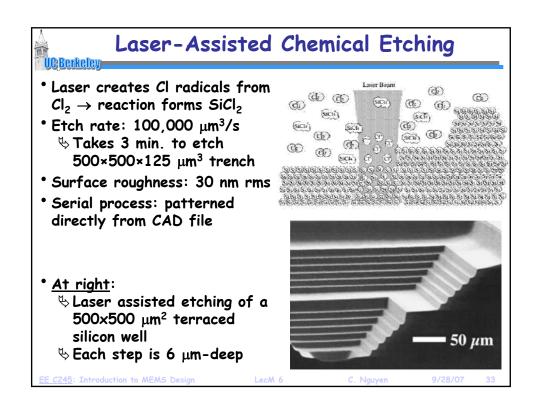
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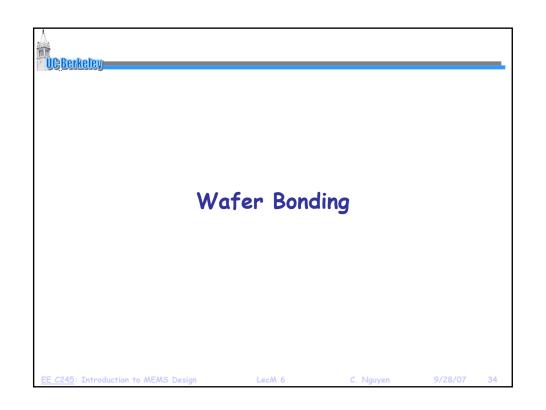


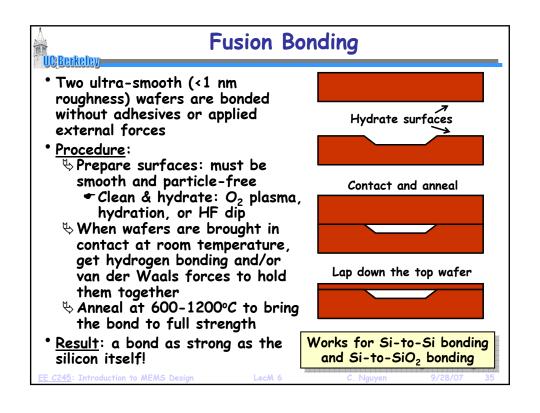


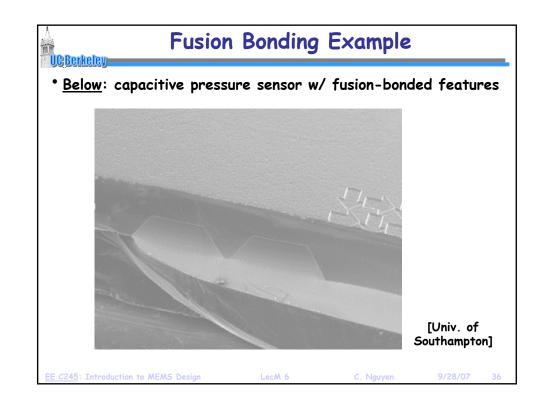


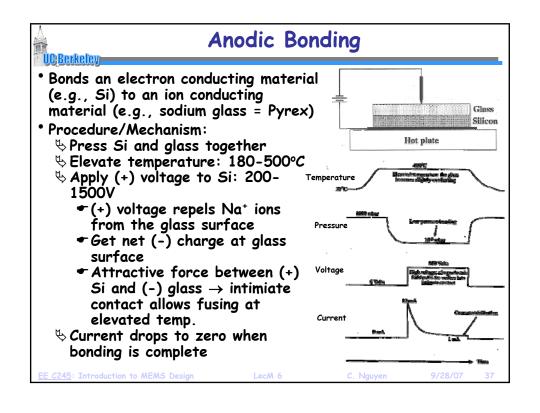
Vapor Phase Etching of Silicon * Vapor phase Xenon Difluoride (XeF2) $2XeF_{2(g)} + Si_{(s)} \rightarrow 2Xe_{(g)} + SiF_{4(g)}$ • Set-up: Xactix XeF₂ ⋄ Xe sublimes at room T Etcher ♥ Closed chamber, 1-4 Torr ♥ Pulsed to control exothermic heat of reaction • Etch rate: 1-3 µm/min, isotropic • Etch masks: photoresist, SiO₂, Si₃N₄, Al, other metals • Issues: ♥ Etched surfaces have granular structure, 10 µm roughness ♦ Hazard: XeF₂ reacts with H₂O in air to form Xe and HF Inductor w/ no substrate [Pister]

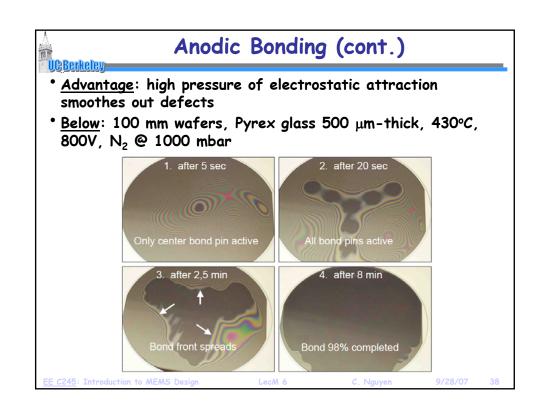












Metal Layer Bonding

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- Pattern seal rings and bond pads photolithographically
- Eutectic bonding
 - Uses eutectic point in metal-Si phase diagrams to form silicides
 - ♦ Au and Si have eutectic point at 363°C
 - ♦ Low temperature process
 - ♦ Can bond slightly rough surfaces
 ♦ Issue: Au contamination of CMOS
- Solder bonding
 - ♦ PbSn (183°C), AuSn (280°C)

 - Scan bond very rough surfaces
 - ♦ Issue: outgassing (not good for encapsulation)
- Thermocompression
 - Scommonly done with electroplated Au or other soft metals
 - ♦ Room temperature to 300°C
 - \$Lowest-T process
 - Scan bond rough surfaces with topography

Thermocompression Bonding • Below: Transfer of hexsil actuator onto CMOS wafer

[Singh, et al, Transducers'97]

