Lecture 23: 11/20/03 A.R. Neureuther

Version Date 11/18/03

EECS 42 Introduction to Digital Electronics

Lecture # 23 Diodes and Diode Circuits

- A) Basic Semiconductor Materials
- **B)** Diode Current and Equation
- C) Diode Circuits

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Motivation

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Digital Circuits, Logic, D/A, etc

- ① We need a "smart switch," i.e., an electronically controlled switch
- ② We need a "gain element" for example, to make comparators.

The device of our dreams exists! \Rightarrow

- kists! ⇒ MOS transistor
- a terrific switch
- low power
- smart

BONUS: MOS is very simple in concept

This week: Basic Semiconductors, Diodes, MOS transistor

Next week: MOS and CMOS Fabrication

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Game Plan

Here is how we begin:

- Learn a little more about semiconductors and pn junction diodes
- 2. Consider the I vs. V model of diodes and their uses in circuits
- 3. Learn about MOSFET Operation as a voltage controlled resistor
- 4. Learn a little about the MOSFET I-V characteristics
- Learn enough about the fabrication process for MOS integrated circuits so that we can visualize the layout of actual CMOS circuits

Thus we begin with a very brief review of semiconductors and doping

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Conductors, Insulators and Semiconductors

Solids with "free electrons" – that is electrons not directly involved in the inter-atomic bonding- are the familiar metals (Cu, Al, Fe, Au, etc).

Solids with no free electrons are the familiar insulators (glass, quartz crystals, ceramics, etc.)

Silicon is an insulator, but at higher temperatures some of the bonding electrons can get free and make it a little conducting – hence the term "semiconductor"

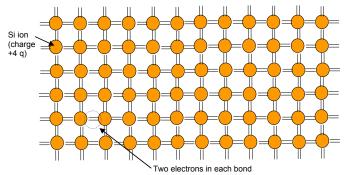
Pure silicon is a poor conductor (and a poor insulator). It has 4 valence electrons, all of which are needed to bond with nearest neighbors. No free electrons.

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Electronic Bonds in Silicon

2-D picture of perfect crystal of pure silicon; double line is a Si-Si bond with each line representing an electron



Actual structure is 3-dimensional tetrahedral- just like carbon bonding in organic and inorganic materials.

Essentially no free electrons, and no conduction →insulator

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How to get conduction in Si?

We must either:

- 1) Chemically modify the Si to produce free carriers (permanent) or
- 2) Electrically "induce" them by the field effect (switchable)

For the first approach controlled impurities, "dopants", are added to Si:

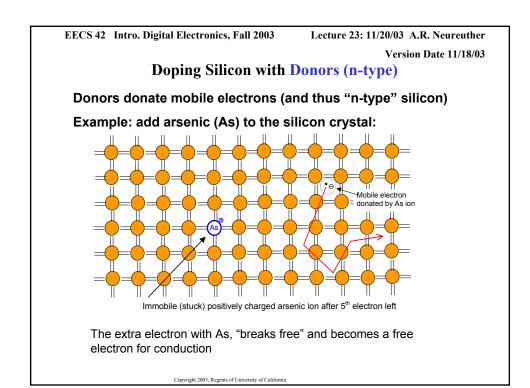
Add group V elements (5 bonding electrons vs four for Si), such as phosphorus or arsenic

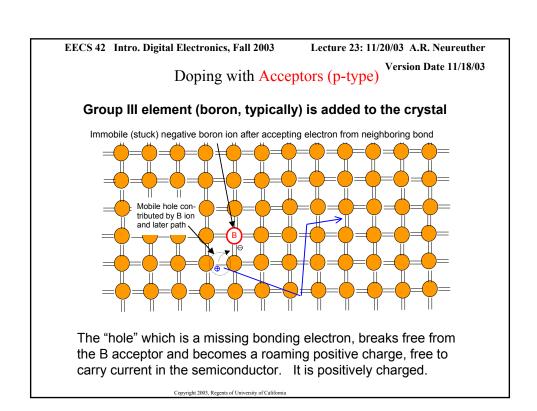
(Extra electrons produce "free electrons" for conduction.)

or

Add group III elements (3 bonding electrons), such as boron

Deficiency of electrons results in "free holes"



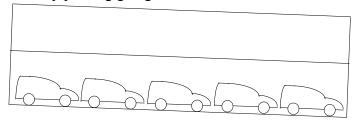


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Shockley's Parking Garage Analogy for Conduction in Si

Two-story parking garage on a hill:



If the lower floor is full and top one is empty, no traffic is possible. Analog of an insulator. All electrons are locked up.

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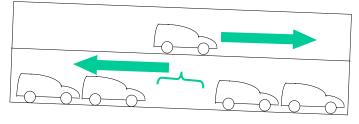
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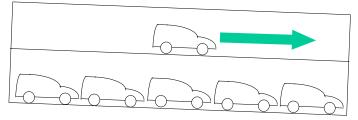
If one car is moved upstairs, it can move AND THE HOLE ON THE LOWER FLOOR CAN MOVE. Conduction is possible. Analog to warmed-up semiconductor. Some electrons get free (and leave "holes" behind).

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Shockley's Parking Garage Analogy for Conduction in Si

Two-story parking garage on a hill:



If an extra car is "donated" to the upper floor, it can move. Conduction is possible. *Analog to N-type semiconductor*. (An electron donor is added to the crystal, creating free electrons).

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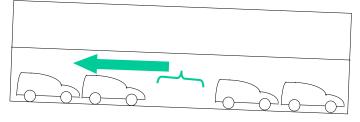
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Shockley's Parking Garage Analogy for Conduction in Si

Two-story parking garage on a hill:



If a car is removed from the lower floor, it leaves a HOLE which can move. Conduction is possible. *Analog to P-type semiconductor*. (Acceptors are added to the crystal, "consuming" bonding electrons, creating free holes).

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Summary of n- and p-type silicon

Pure silicon is an insulator. At high temperatures it conducts weakly.

If we add an impurity with extra electrons (e.g. arsenic, phosphorus) these extra electrons are set free and we have a pretty good conductor (n-type silicon).

If we add an impurity with a deficit of electrons (e.g. boron) then bonding electrons are missing (holes), and the resulting holes can move around ... again a pretty good conductor (p-type silicon)

Now what is really interesting is when we join n-type and p-type silicon, that is make a pn junction. It has interesting electrical properties.

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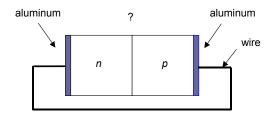
Junctions of n- and p-type Regions

p-n junctions form the essential basis of all semiconductor devices.

A silicon chip may have 10⁸ to 10⁹ p-n junctions today.

How do they behave*? What happens to the electrons and holes? What is the electrical circuit model for such junctions?

n and p regions are brought into contact:



^{*}Note that the textbook has a very good explanation.

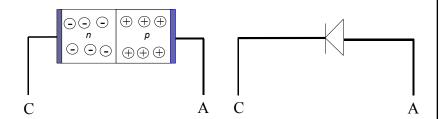


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A pn junction is formed - what happens?

The structure and the circuit symbol are shown below:



The electrons are depicted as \bigcirc . Note that the n-type silicon is actually electrically neutral, but we emphasize the "free" electrons...

The holes in the p-type silicon are depicted as \oplus . Again, the material is electrically neutral.

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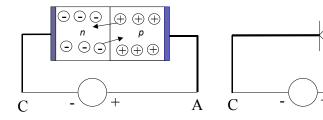
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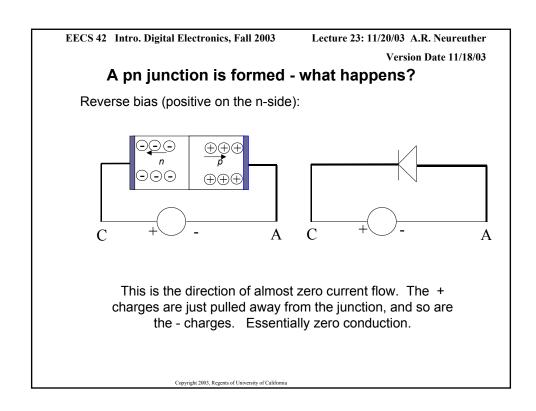
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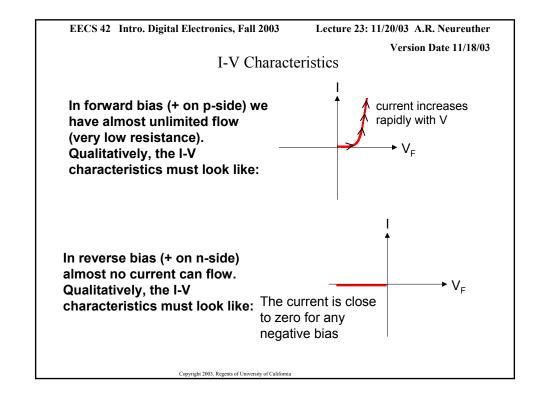
A pn junction is formed - what happens?

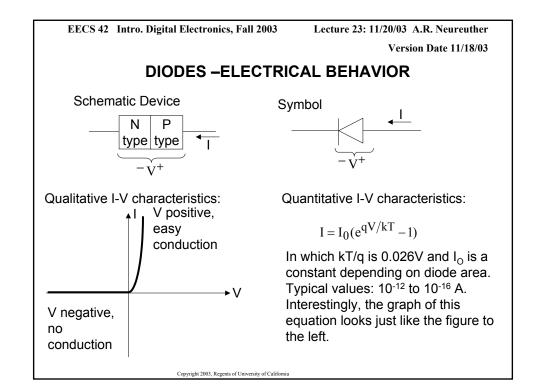
Forward bias (positive on the p-side):



This is the direction of easy current flow. + charges flow to meet up with – charges. Essentially unlimited conduction.







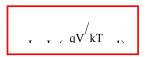
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THE PN JUNCTION DIODE I vs. V

I-V characteristic of PN junctions

In EECS 105, 130, and other courses you will learn why the I vs. V relationship for PN junctions is of the form



where I_0 is a constant proportional to junction area and depending on doping in P and N regions, k is Boltzman constant, and T is absolute temperature.

We note that in forward bias, I increases **exponentially** and is in the μ A-mA range for voltages typically in the range of 0.6-0.8V. In reverse bias, the current is essentially zero.

