EECS 42 Intro. Digital Electronics, Fall 2003

Lecture 23: 11/20/03 A.R. Neureuther

Version Date 11/18/03

EECS 42 Introduction to Digital Electronics

Lecture # 23 Diodes and Diode Circuits

- A) Basic Semiconductor Materials
- **B)** Diode Current and Equation
- C) Diode Circuits

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EECS 42 Intro. Digital Electronics, Fall 2003 Lecture 23: 11/20/03 A.R. Neureuther Version Date 11/18/03 Motivation ① We need a "smart switch," i.e., an electronically controlled switch Digital Circuits, Logic, D/A, etc ② We need a "gain element" – for example, to make comparators. The device of our dreams exists! = **MOS** transistor · a terrific switch · low power smart BONUS: MOS is very simple in concept This week: Basic Semiconductors, Diodes, MOS transistor Next week: MOS and CMOS Fabrication

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Game Plan

Here is how we begin:

- Learn a little more about semiconductors and pn junction diodes
- 2. Consider the I vs. V model of diodes and their uses in circuits
- 3. Learn about MOSFET Operation as a voltage controlled resistor
- 4. Learn a little about the MOSFET I-V characteristics
- Learn enough about the fabrication process for MOS integrated circuits so that we can visualize the layout of actual CMOS circuits

Thus we begin with a very brief review of semiconductors and doping

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Conductors, Insulators and Semiconductors

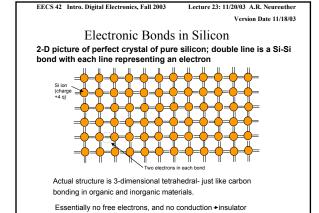
Solids with "free electrons" – that is electrons not directly involved in the inter-atomic bonding- are the familiar metals (Cu, AI, Fe, Au, etc).

Solids with no free electrons are the familiar insulators (glass, quartz crystals, ceramics, etc.)

Silicon is an insulator, but at higher temperatures some of the bonding electrons can get free and make it a little conducting – hence the term "semiconductor"

Pure silicon is a poor conductor (and a poor insulator). It has 4 valence electrons, all of which are needed to bond with nearest neighbors. No free electrons.

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How to get conduction in Si?

We must either:

1) Chemically modify the Si to produce free carriers (permanent) or

2) Electrically "induce" them by the field effect (switchable)

For the first approach controlled impurities, "dopants", are added to Si:

Si:

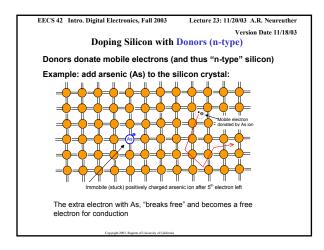
Add group V elements (5 bonding electrons vs four for Si), such as phosphorus or arsenic

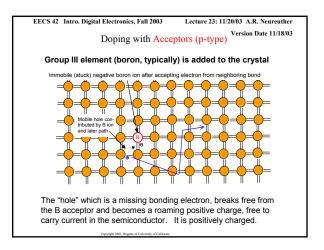
(Extra electrons produce "free electrons" for conduction.)

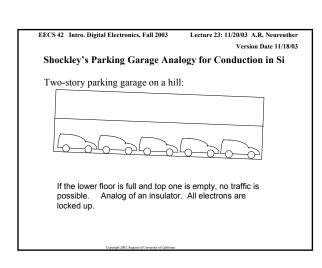
or

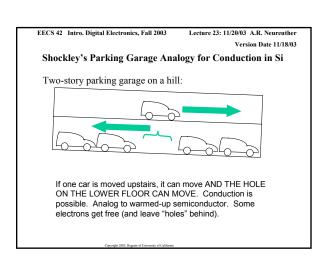
Add group III elements (3 bonding electrons), such as boron

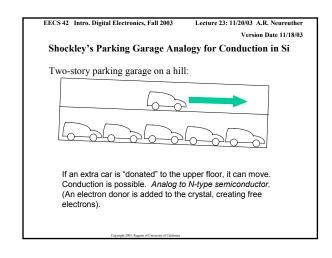
Deficiency of electrons results in "free holes"

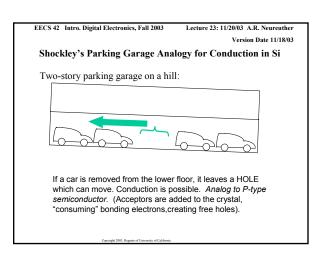












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Summary of n- and p-type silicon

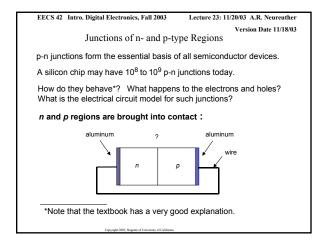
Pure silicon is an insulator. At high temperatures it conducts weakly.

If we add an impurity with extra electrons (e.g. arsenic, phosphorus) these extra electrons are set free and we have a pretty good conductor (n-type silicon).

If we add an impurity with a deficit of electrons (e.g. boron) then bonding electrons are missing (holes), and the resulting holes can move around ... again a pretty good conductor (p-type silicon)

Now what is really interesting is when we join n-type and p-type silicon, that is make a pn junction. It has interesting electrical properties.

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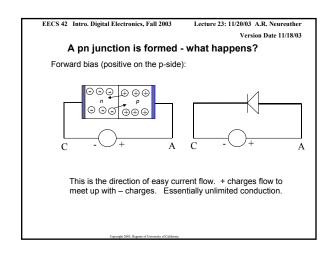
A pn junction is formed - what happens?

The structure and the circuit symbol are shown below:

C A C A

The electrons are depicted as ③. Note that the n-type silicon is actually electrically neutral, but we emphasize the "free" electrons..

The holes in the p-type silicon are depicted as ⊕. Again, the material is electrically neutral.



Reverse bias (positive on the n-side):

This is the direction of almost zero current flow. The + charges are just pulled away from the junction, and so are the - charges. Essentially zero conduction.

