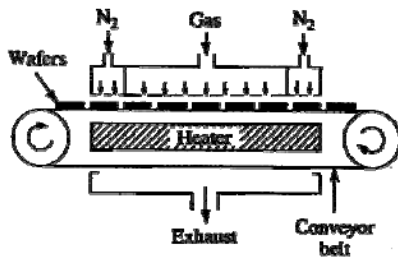


## Atmospheric Pressure Reactor (APCVD)

UC Berkeley



- Once used for silicon dioxide passivation in integrated circuits
- Substrates fed continuously
- Large diameter wafers
- Need high gas flow rates
- Mass transport-limited regime (high pressure, so tougher for gas to get to the wafer surface)

### Problems/Issues:

- ↳ Wafers lay flat, and thus, incorporate foreign particles
- ↳ Poor step coverage

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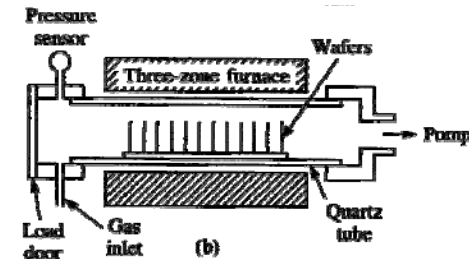
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## Low Pressure Reactor (LPCVD)

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- Many films available: polysilicon, SiGe, Si<sub>3</sub>N<sub>4</sub>, SiO<sub>2</sub>, phosphosilicate glass (PSG), BPSG, W

- Temp.: 300 → 1150°C
- Press.: 30 → 250 Pa (200mTorr → 2Torr)
- Reaction rate limited; reduced pressure gives gas molecular high diffusivity; can supply reactants very fast!
- Can handle several hundred wafers at a time
- Excellent uniformity



### Problems:

- ↳ Low dep. rate (compared to atm.)
- ↳ Higher T (than atmospheric)
- ↳ In hot wall reactors, get deposition on tube walls (must clean)

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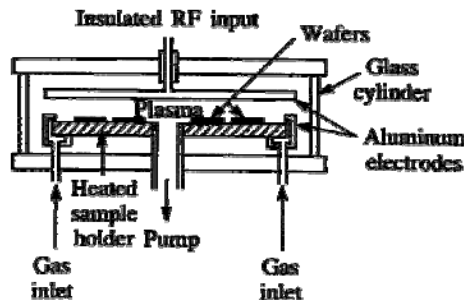
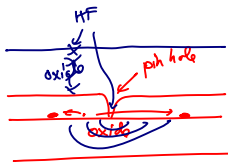
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## Plasma-Enhanced CVD Reactor (PECVD)

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- RF-induced glow discharge + thermal energy to drive reactions → allows lower temperature deposition with decent conformability
- Still low pressure



### Problems:

- ↳ Pin-holes
- ↳ Non-stoichiometric films
- ↳ Incorporation of H<sub>2</sub>, N<sub>2</sub>, O<sub>2</sub> contaminants in film; can lead to outgassing or bubbling in later steps

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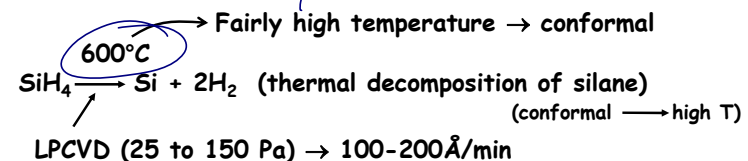
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## Polysilicon CVD

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### Polysilicon Deposition:



### In situ doping of polysilicon:

- ↳ n-type: add PH<sub>3</sub> (phosphine) or Arsine gases (but greatly reduces dep. rate)
- ↳ p-type: add diborane gas (greatly increases dep. Rate)


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## Silicon Oxide CVD

**Silicon Dioxide Deposition:**

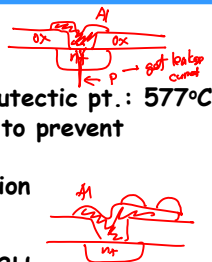
- **After metallization** (e.g., over aluminum)
  - ↳ Temperature cannot exceed the Si-Al eutectic pt.:  $577^{\circ}\text{C}$
  - ↳ Actually, need lower than this ( $<500^{\circ}\text{C}$ ) to prevent hillocks from growing on Al surfaces
  - ↳ Similar issues for copper (Cu) metallization
- Low temperature reactions:
 

LPCVD  
LTO  
Reactions

{

$$\begin{array}{l} \text{SiH}_4 + \text{O}_2 \xrightarrow{300-500^{\circ}\text{C}} \text{SiO}_2 + 2\text{H}_2 \\ \text{(silane)} \\ \\ 4\text{PH}_3 + 5\text{O}_2 \xrightarrow{300-500^{\circ}\text{C}} 2\text{P}_2\text{O}_5 + 6\text{H}_2 \\ \text{(phosphine)} \end{array}$$

Phosphosilicate  
glass (PSG)
- Above reactions: not very conformal step coverage  $\rightarrow$  need higher T for this



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