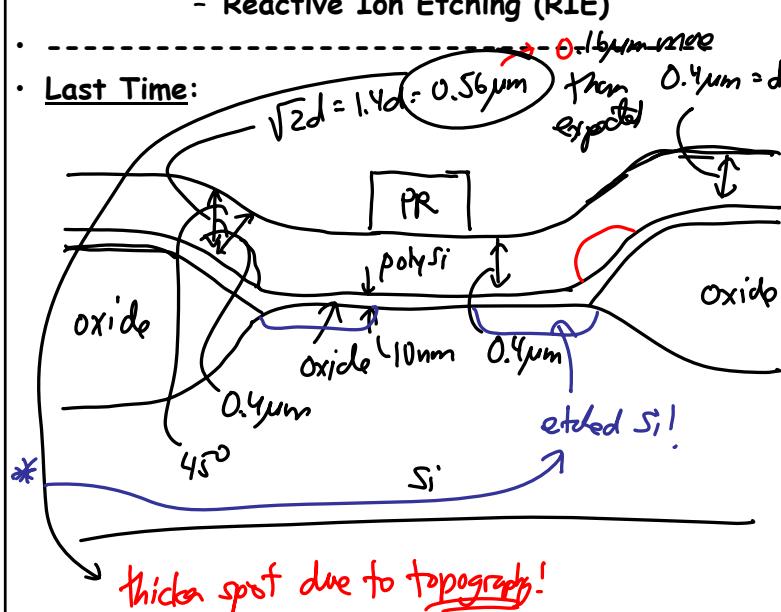


Lecture 14c: Etching I

Lecture 14: Etching

- Announcements:
- Might need volunteers for Cal Day: if interested, sign up with your lab TA
- Midterm Exam: coming Thursday, March 18
 - Evening exam? 9 for different time; 14 for this period → so it'll be during lecture
- Lecture Topics:
 - Etching
 - Anisotropy
 - Selectivity
 - Deep Reactive Ion Etching (DRIE)
 - Wet Etching
 - Dry Etching
 - Plasma Etching
 - Reactive Ion Etching (RIE)



↓
Thus, must overetch by at least 40%:
 $40\% \text{ overetch} \rightarrow (0.4)(0.4) = 0.16\mu m \text{ poly-Si}$
= ?? oxide

Depends upon the selectivity of the etchant to
polysilicon versus oxide

Define selectivity to A over B:

$$S_{ab} = \frac{E.R._a}{E.R._b} \quad \begin{matrix} \leftarrow \text{etch rate of } A \\ \leftarrow \text{etch rate of } B \\ \curvearrowright \text{selectivity of } a \text{ over } b \end{matrix}$$

e.g., wet polysilicon etch ($HNO_3 + NH_4F + H_2O$)

$$S_{\text{poly/SiO}_2} = \frac{15}{1} \quad (\text{very good selectivity})$$

$S_{\text{poly/PR}} = \text{very high}$

(but PR can still peel off after soaking
 $f_a > 30 \text{ min.}, \text{ so beware!}$)

Lecture 14c: Etching I

e.g., polysilicon dry etch:
regular RIE

$$S_{\text{poly}/\text{SiO}_2} = \frac{5-7}{1} \quad (\text{but depends on the type of etcher})$$

very high density plasma → ECR: 30:1
Bosch: 100:1 (a better)

If $S_{\text{poly}/\text{ox}} = \frac{8}{1} \Rightarrow 40\%$ oxide removes:

$\frac{0.16\mu}{8} = 20\text{ nm of oxide!}$
 ↓ this will etch all the polysilicon over the SiO_2 in the active area;
 then etch all the SiO_2 (10nm);
 then start etching the S/D drain regions very fast! Bad

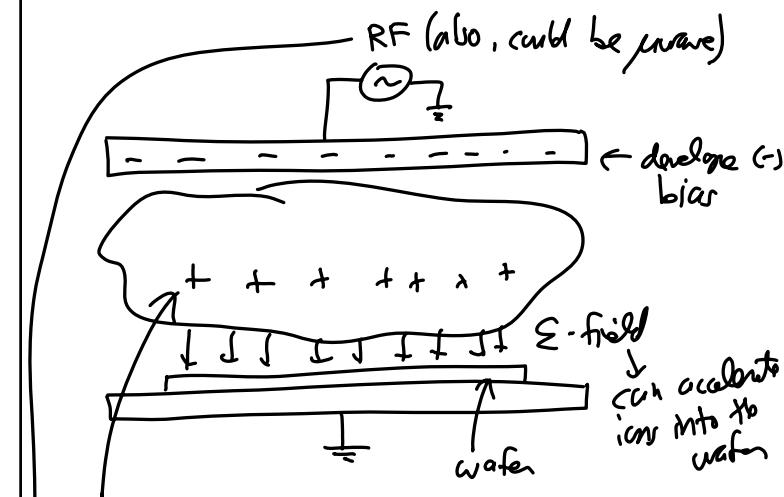
or better selectivity,
e.g., $S_{\text{poly}/\text{SiO}_2} = \frac{20}{1}$ (ECR)
high density plasma

$$\frac{40\% \text{ oxide removes}}{30} \frac{0.16}{30} = 5.3\text{ nm (better)}$$

Go through module for wet etching.

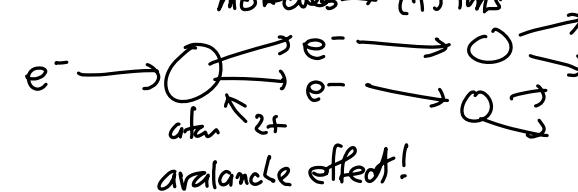
Dry Etching

- Physical Sputtering
 - Plasma Etching
 - Reactive Ion Etching
- } All based upon plasma processes!



plasma (partially ionized gas composed of ions, e^- , and highly reactive neutral species)

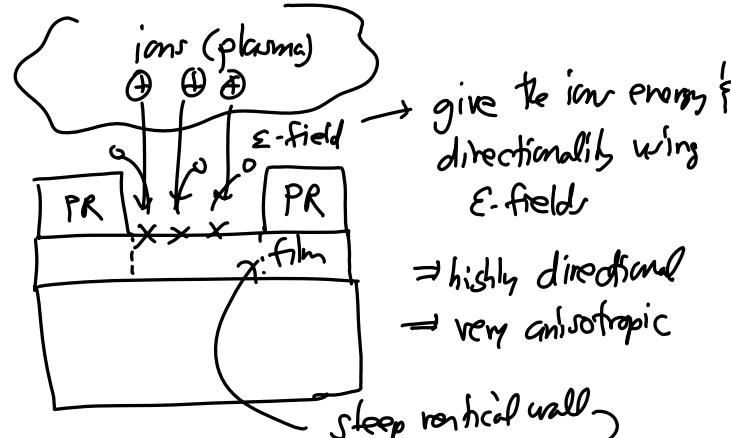
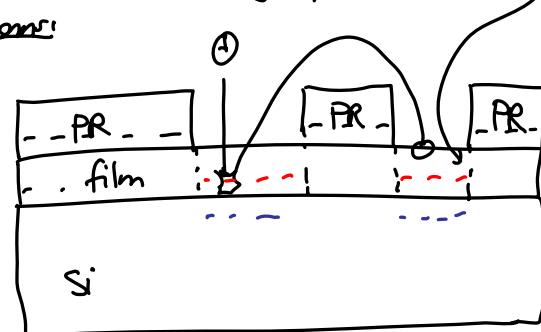
RF energy → inelastic collision between e^- 's & molecules → (f) ions



avalanche effect!

Lecture 14c: Etching IPhysical Sputtering (ion milling)

→ bombard the substrate w/ energetic ions
↳ etching via physical momentum transfer

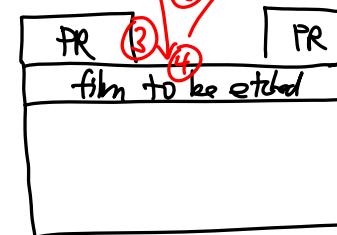
Problems:

- ① PR a often masking material etched at almost the same rate as the film to be etched (very poor selectivity!)
- ② Ejected species not inherently volatile
↳ get redeposition → non-uniform etch!

Plasma Etching

→ plasma (gas glow discharge) creates reactive species that chemically react w/ the film to be etched

→ result: much better selectivity, but got an

plasma isotropic etchPlasma Etching Mechanism

- ① Reactive species generated in a plasma.
- ② " " " diffuse to the surface of the material to be etched
- ③ Species adsorb on the surface.
- ④ Chemical reaction.

Lecture 14c: Etching I

- ⑤ By-product desorbed from surface.
- ⑥ Desorbed species must diffuse into the gas stream.

MOST IMPORTANT STEP!
(determines whether plasma etch is possible or not)