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## EE C247B - ME C218 Introduction to MEMS Design Spring 2017

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Lecture Module 6: Bulk Micromachining

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## Lecture Outline

- Reading: Senturia Chpt. 3, Jaeger Chpt. 11, Handouts: "Bulk Micromachining of Silicon"
- Lecture Topics:
  - ↳ Bulk Micromachining
  - ↳ Anisotropic Etching of Silicon
  - ↳ Boron-Doped Etch Stop
  - ↳ Electrochemical Etch Stop
  - ↳ Isotropic Etching of Silicon
  - ↳ Deep Reactive Ion Etching (DRIE)

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## Bulk Micromachining

- Basically, etching the substrate (usually silicon) to achieve microstructures
- Etching modes:
  - ↳ Isotropic vs. anisotropic
  - ↳ Reaction-limited
    - Etch rate dep. on temp.
  - ↳ Diffusion-limited
    - Etch rate dep. on mixing
    - Also dependent on layout & geometry, i.e., on loading
- Choose etch mode based on
  - ↳ Desired shape
  - ↳ Etch depth and uniformity
  - ↳ Surface roughness (e.g., sidewall roughness after etching)
  - ↳ Process compatibility (w/ existing layers)
  - ↳ Safety, cost, availability, environmental impact

	Wet etch	Plasma (dry) etch
Isotropic		
Anisotropic		

adsorption → surface reaction → desorption

slowest step controls rate of reaction

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## Mechanical Properties of Silicon

- Crystalline silicon is a hard and brittle material that deforms elastically until it reaches its yield strength, at which point it breaks.
  - ↳ Tensile yield strength = 7 GPa (~1500 lb suspended from 1 mm<sup>2</sup>)
  - ↳ Young's Modulus near that of stainless steel
  - ↳ {100} = 130 GPa; {110} = 169 GPa; {111} = 188 GPa
  - ↳ Mechanical properties uniform, no intrinsic stress
  - ↳ Mechanical integrity up to 500°C
  - ↳ Good thermal conductor
  - ↳ Low thermal expansion coefficient
  - ↳ High piezoresistivity

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### Anisotropic Etching of Silicon

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- Etching of Si w/ KOH
 
$$\text{Si} + 2\text{OH}^- \rightarrow \text{Si}(\text{OH})_2^{2-} + 4\text{e}^-$$

$$4\text{H}_2\text{O} + 4\text{e}^- \rightarrow 4(\text{OH})^- + 2\text{H}_2$$
- Crystal orientation dependent etch rates
  - {110}:{100}:{111}=600:400:1
  - {100} and {110} have 2 bonds below the surface & 2 dangling bonds that can react
  - {111} plane has three of its bonds below the surface & only one dangling bond to react → much slower E.R.
  - {111} forms protective oxide
  - {111} smoother than other crystal planes → good for optical MEMS (mirrors)

Self-limiting etches

Membrane

Front side mask

Back side mask

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### Anisotropic Etching of Silicon

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- Deposit nitride:
  - Target = 100nm
  - 22 min. LPCVD @ 800°C
- Lithography to define areas of silicon to be etched
- Etch/pattern nitride mask
  - RIE using  $\text{SF}_6$
  - Remove PR in PRS2000
- Etch the silicon
  - Use 1:2 KOH:H<sub>2</sub>O (wt.), stirred bath @ 80°C
  - Etch Rates:
    - (100) Si → 1.4 μm/min
    - Si<sub>3</sub>N<sub>4</sub> → ~ 0 nm/min
    - SiO<sub>2</sub> → 1-10 nm/min
    - Photoresist, Al → fast
- Micromasking by H<sub>2</sub> bubbles leads to roughness
  - Stir well to displace bubbles
  - Can also use oxidizer for (111) surfaces
  - Or surfactant additives to suppress bubble formation

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### Silicon Wafers

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{110} plane

{100} planes

{100} plane

45°

{110} primary flat

{100} type wafer

[Maluf]

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### Silicon Crystallography

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[001] z

[110] y

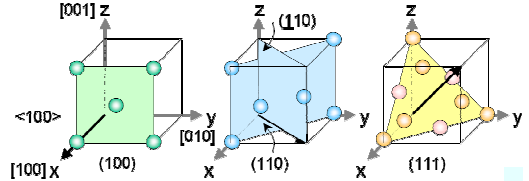
[111] x

Miller Indices (h k l):

- Planes
  - Reciprocal of plane intercepts with axes
  - e.g., for (110), intercepts: (x,y,z) = (1,1,∞); reciprocals: (1,1,0) → (110)
  - (unique), {family}
- Directions
  - One endpoint of vector @ origin
  - [unique], <family>

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### Determining Angles Between Planes

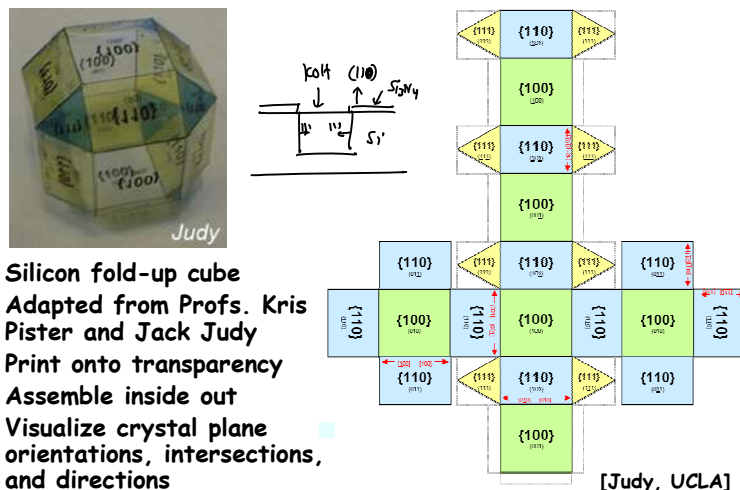


- The angle between vectors  $[abc]$  and  $[xyz]$  is given by:
$$\cos \theta = \frac{ax + by + cz}{|(a, b, c)| \cdot |(x, y, z)|}$$

$$\theta_{(a,b,c),(x,y,z)} = \cos^{-1} \left[ \frac{ax + by + cz}{|(a, b, c)| \cdot |(x, y, z)|} \right]$$
- For  $\{100\}$  and  $\{110\} \rightarrow 45^\circ$
- For  $\{100\}$  and  $\{111\} \rightarrow 54.74^\circ$
- For  $\{110\}$  and  $\{111\} \rightarrow 35.26^\circ, 90^\circ, \text{ and } 144.74^\circ$

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### Silicon Crystal Origami



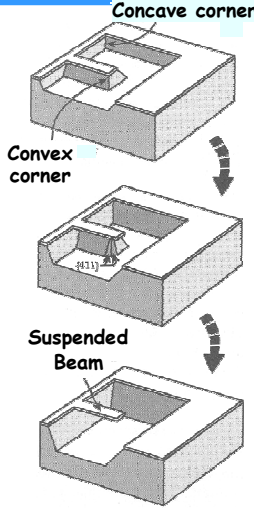
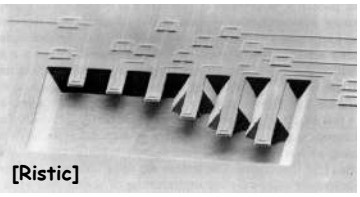
- Silicon fold-up cube
- Adapted from Profs. Kris Pister and Jack Judy
- Print onto transparency
- Assemble inside out
- Visualize crystal plane orientations, intersections, and directions

[Judy, UCLA]

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### Undercutting Via Anisotropic Si Etching

- Concave corners bounded by  $\{111\}$  are not attacked
- ... but convex corners bounded by  $\{111\}$  are attacked
  - Two  $\{111\}$  planes intersecting now present two dangling bonds  $\rightarrow$  no longer have just one dangling bond  $\rightarrow$  etch rate fast
  - Result: can undercut regions around convex corners

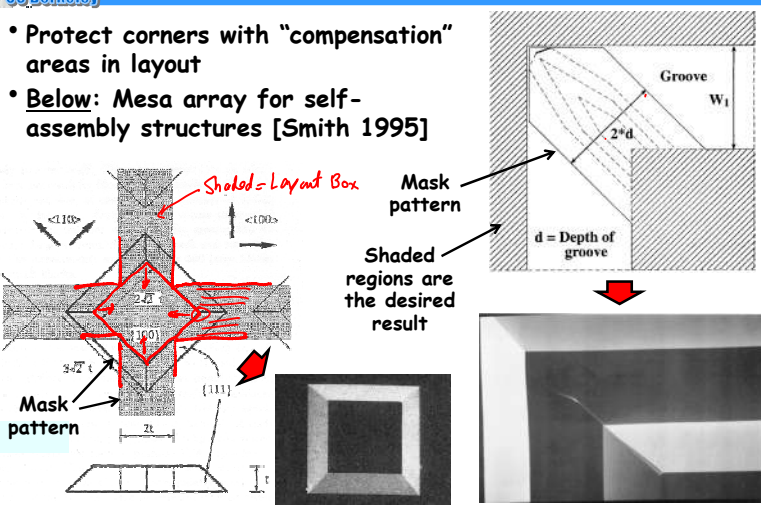



[Ristic]

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### Corner Compensation

- Protect corners with "compensation" areas in layout
- Below: Mesa array for self-assembly structures [Smith 1995]



Shaded regions are the desired result

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### Other Anisotropic Silicon Etchants

- TMAH, Tetramethyl ammonium hydroxide, 10-40 wt.% (90°C)
  - ↳ Etch rate (100) = 0.5-1.5  $\mu\text{m}/\text{min}$
  - ↳ Attacks Al
    - Si-doped Al safe & IC compatible
  - ↳ Etch ratio (100)/(111) = 10-35
  - ↳ Etch masks:  $\text{SiO}_2$ ,  $\text{Si}_3\text{N}_4$  ~ 0.05-0.25 nm/min
  - ↳ Boron doped etch stop, up to 40 $\times$  slower
- EDP (115°C)
  - ↳ Carcinogenic, corrosive
  - ↳ Etch rate (100) = 0.75  $\mu\text{m}/\text{min}$
  - ↳ Al may be etched
  - ↳  $R(100) > R(110) > R(111)$
  - ↳ Etch ratio (100)/(111) = 35
  - ↳ Etch masks:  $\text{SiO}_2$  ~ 0.2 nm/min,  $\text{Si}_3\text{N}_4$  ~ 0.1 nm/min
  - ↳ Boron doped etch stop, 50 $\times$  slower

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### Boron-Doped Etch Stop

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### Boron-Doped Etch Stop

- Control etch depth precisely with boron doping (p++)
  - ↳  $[\text{B}] > 10^{20} \text{ cm}^{-3}$  reduces KOH etch rate by 20-100 $\times$
  - ↳ Can use gaseous or solid boron diffusion
  - ↳ Recall etch chemistry:
 
$$\text{Si} + 2\text{OH}^- \rightarrow \text{Si}(\text{OH})_2^{2+} + 4\text{e}^-$$

$$4\text{H}_2\text{O} + 4\text{e}^- \rightarrow 4(\text{OH})^- + 2\text{H}_2$$
  - ↳ At high dopant levels, injected electrons recombine with holes in valence band and are unavailable for reactions to give  $\text{OH}^-$
- Result:
  - ↳ Beams, suspended films
  - ↳ 1-20  $\mu\text{m}$  layers possible

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### Ex: Micronozzle

- Micronozzle using anisotropic etch-based fabrication
- Used for inkjet printer heads

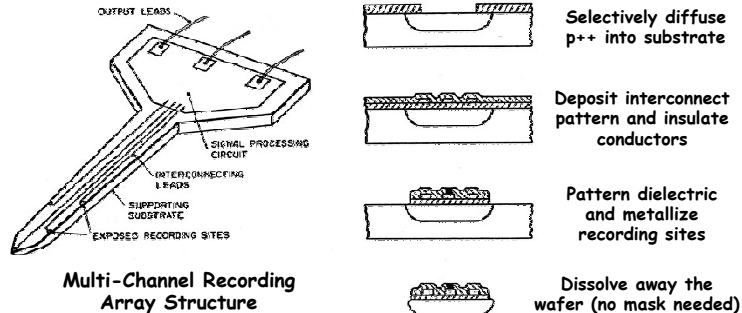
[Maluf]

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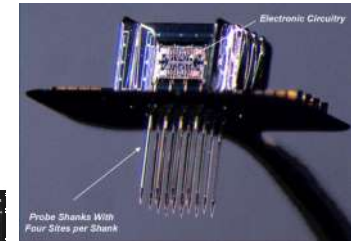
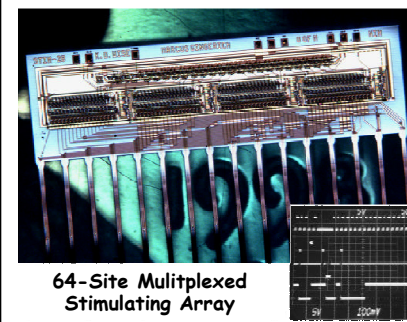
### Ex: Microneedle

- Below: micro-neurostimulator
  - Used to access central nervous system tissue (e.g., brain) and record electrical signals on a cellular scale
- Wise Group, Univ. of Michigan



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### Ex: Microneedles (cont.)



[Wise, U. of Michigan]

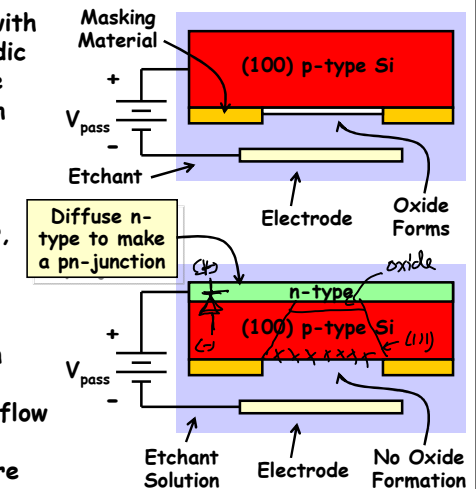
- Micromachined with on-chip CMOS electronics
- Both stimulation and recording modes
- 400  $\mu\text{m}$  site separations, extendable to 3D arrays
- Could be key to neural prosthesis systems focusing on the central nervous system

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### Electrochemical Etch Stop

### Electrochemical Etch Stop

- When silicon is biased with a sufficiently large anodic potential relative to the etchant  $\rightarrow$  get oxidation (i.e., electrochemical passivation), which then prevents etching
- For passivation to occur, current flow is required
- If current flow can be prevented  $\rightarrow$  no oxide growth, and etching can proceed
  - Can prevent current flow by adding a reverse-biased diode structure



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### Electrochemical Etch Stop

- Electrochemical etch stop
  - n-type epitaxial layer grown on p-type wafer forms p-n junction diode
    - $V_p > V_n \rightarrow$  electrical conduction (current flow) *→ form oxide → prevents etching*
    - $V_p < V_n \rightarrow$  reverse bias current (very little current flow) *→ prevent oxide formation → allow etching*
- Passivation potential: potential at which thin  $\text{SiO}_2$  film forms
  - different for p-Si and n-Si, but basically need the Si to be the anode in an electrolytic setup
- Setup:
  - p-n diode in reverse bias
  - p-substrate floating  $\rightarrow$  etched
  - n-layer above passivation potential  $\rightarrow$  not etched

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### Electrochemical Etching of CMOS

- N-type Si well with circuits suspended f/  $\text{SiO}_2$  support beam
- Thermally and electrically isolated
- If use TMAH etchant, doped (w/Si) Al bond pads safe

[Reay, et al. (1994)]  
[Kovacs Group, Stanford]

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### Ex: Bulk Micromachined Pressure Sensors

- Piezoresistivity: change in electrical resistance due to mechanical stress
- In response to pressure load on thin Si film, piezoresistive elements change resistance
- Membrane deflection  $< 1 \mu\text{m}$

[Maluf]

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### Ex: Pressure Sensors

- Below: catheter tip pressure sensor [Lucas NovaSensor]
  - Only  $150 \times 400 \times 900 \mu\text{m}^3$

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### Deep Reactive-Ion Etching (DRIE)

**The Bosch process:**

- Inductively-coupled plasma
- Etch Rate: 1.5-4  $\mu\text{m}/\text{min}$
- Two main cycles in the etch:
  - Etch cycle** (5-15 s):  $\text{SF}_6$  ( $\text{SF}_x^+$ ) etches Si
  - Deposition cycle** (5-15 s):  $\text{C}_4\text{F}_8$  deposits fluorocarbon protective polymer ( $\text{CF}_2$ )<sub>n</sub>
- Etch mask selectivity:
  - $\text{SiO}_2 \sim 200:1$
  - Photoresist  $\sim 100:1$
- Issue:** finite sidewall roughness
  - scalloping < 50 nm
- Sidewall angle:  $90^\circ \pm 2^\circ$

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### DRIE Issues: Etch Rate Variance

- Etch rate is diffusion-limited and drops for narrow trenches
  - Adjust mask layout to eliminate large disparities
  - Adjust process parameters (slow down the etch rate to that governed by the slowest feature)

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### DRIE Issues: "Footing"

- Etch depth precision
  - Etch stop: buried layer of  $\text{SiO}_2$
  - Due to 200:1 selectivity, the (vertical) etch practically just stops when it reaches  $\text{SiO}_2$
- Problem:** Lateral undercut at Si/ $\text{SiO}_2$  interface → "footing"
  - Caused by charge accumulation at the insulator

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### Recipe-Based Suppression of "Footing"

- Use **higher process pressure** to reduce ion charging [Nozawa]
  - High operating pressure → concentration of (-) charge increases and can neutralize (+) surface charge
  - Issue:** must introduce as a separate recipe when the etch reaches the Si-insulator interface, so must be able to very accurately predict the time needed for etching
- Adjust etch recipe** to reduce overetching [Schmidt]
  - Change  $\text{C}_4\text{F}_8$  flow rate, pressure, etc., to enhance passivation and reduce overetching
  - Issue:** Difficult to simultaneously control footing in a narrow trench and prevent grass in wide trenches
- Use **lower frequency plasma** to avoid surface charging [Morioka]
  - Low frequency → more ions with low directionality and kinetic energy → neutralizes (-) potential barrier at trench entrance
  - Allows e<sup>-</sup>s to reach the trench base and neutralize (+) charge → maintain charge balance inside the trench

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### Metal Interlayer to Prevent "Footing"

Pre-defined metal interlayer grounded to substrate supplies e's to neutralize (+) charge and prevent charge accumulation at the Si-insulator interface

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### Footing Prevention (cont.)

- Below: DRIE footing over an oxide stop layer
- Right: efficacy of the metal interlayer footing prevention approach [Kim, Stanford]

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### DRIE Examples

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### Vapor Phase Etching of Silicon

- Vapor phase Xenon Difluoride ( $\text{XeF}_2$ )  

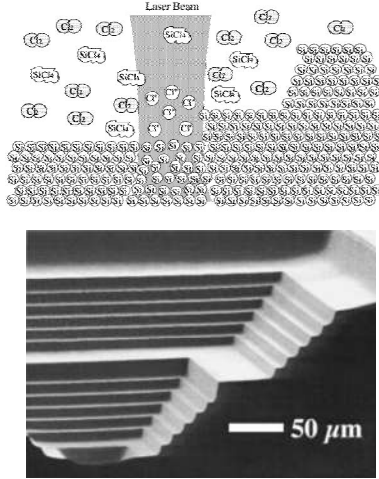
$$2\text{XeF}_{2(g)} + \text{Si}_{(s)} \rightarrow 2\text{Xe}_{(g)} + \text{SiF}_{4(g)}$$
- Set-up:
  - Xe sublimates at room T
  - Closed chamber, 1-4 Torr
  - Pulsed to control exothermic heat of reaction
- Etch rate: 1-3  $\mu\text{m}/\text{min}$ , isotropic
- Etch masks: photoresist,  $\text{SiO}_2$ ,  $\text{Si}_3\text{N}_4$ , Al, other metals
- Issues:
  - Etched surfaces have granular structure, 10  $\mu\text{m}$  roughness
  - Hazard:  $\text{XeF}_2$  reacts with  $\text{H}_2\text{O}$  in air to form Xe and HF

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### Laser-Assisted Chemical Etching

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- Laser creates Cl radicals from  $\text{Cl}_2 \rightarrow$  reaction forms  $\text{SiCl}_2$
- Etch rate:  $100,000 \mu\text{m}^3/\text{s}$ 
  - Takes 3 min. to etch  $500 \times 500 \times 125 \mu\text{m}^3$  trench
- Surface roughness: 30 nm rms
- Serial process: patterned directly from CAD file
- At right:
  - Laser assisted etching of a  $500 \times 500 \mu\text{m}^2$  terraced silicon well
  - Each step is 6  $\mu\text{m}$ -deep



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### Wafer Bonding

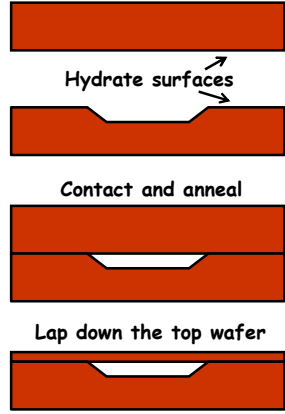
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### Fusion Bonding

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- Two ultra-smooth ( $<1 \text{ nm}$  roughness) wafers are bonded without adhesives or applied external forces
- Procedure:
  - Prepare surfaces: must be smooth and particle-free
    - Clean & hydrate:  $\text{O}_2$  plasma, hydration, or HF dip
  - When wafers are brought in contact at room temperature, get hydrogen bonding and/or van der Waals forces to hold them together
  - Anneal at  $600\text{--}1200^\circ\text{C}$  to bring the bond to full strength
- Result: a bond as strong as the silicon itself!



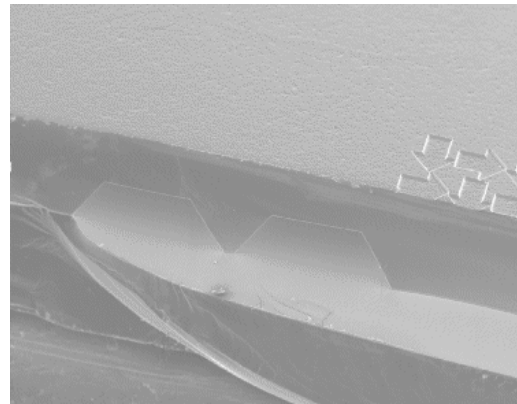
Works for Si-to-Si bonding and Si-to- $\text{SiO}_2$  bonding

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### Fusion Bonding Example

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- Below: capacitive pressure sensor w/ fusion-bonded features



[Univ. of Southampton]

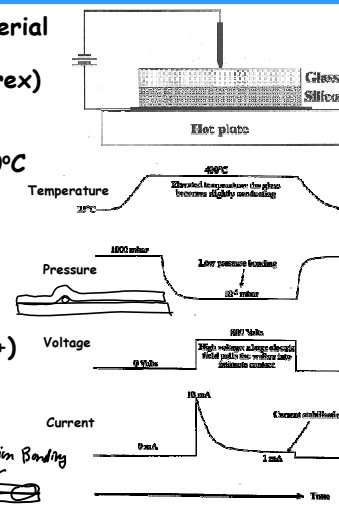
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### Anodic Bonding

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- Bonds an electron conducting material (e.g., Si) to an ion conducting material (e.g., sodium glass = Pyrex)
- Procedure/Mechanism:
  - Press Si and glass together
  - Elevate temperature: 180-500°C
  - Apply (+) voltage to Si: 200-1500V
    - (+) voltage repels  $\text{Na}^+$  ions from the glass surface
    - Get net (-) charge at glass surface
    - Attractive force between (+) Si and (-) glass  $\rightarrow$  intimate contact allows fusing at elevated temp.
  - Current drops to zero when fusion bonding is complete



The diagram shows a cross-section of a silicon wafer on a glass substrate, with an electric plate below. To the right, four graphs show the process parameters over time:
 

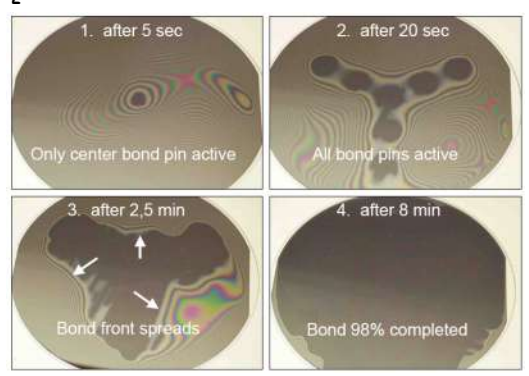
- Temperature:** Starts at 25°C, rises to 450°C, and then levels off. A note indicates: 'Elevated temperature: the glass becomes slightly conducting'.
- Pressure:** Starts at 1000 mbar, drops to 10<sup>-4</sup> mbar, and then rises back to 1000 mbar. A note indicates: 'Lower pressure bonding'.
- Voltage:** Starts at 0V, rises to 800V, and then drops to 0V. A note indicates: 'High voltage along electrical field pulls the wafer into intimate contact'.
- Current:** Starts at 0mA, rises to 10mA, and then drops to 0mA. A note indicates: 'Current nullification'.

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### Anodic Bonding (cont.)

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- Advantage:** high pressure of electrostatic attraction smooths out defects
- Below:** 100 mm wafers, Pyrex glass 500  $\mu\text{m}$ -thick, 430°C, 800V,  $\text{N}_2$  @ 1000 mbar



The four images show the progression of anodic bonding on a circular wafer:
 

1. after 5 sec: Only center bond pin active
2. after 20 sec: All bond pins active
3. after 2.5 min: Bond front spreads
4. after 8 min: Bond 98% completed

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