

Lecture 8: Surface Micromachining II

- Announcements:
- HW#2 due Tuesday, 2/25, at 8 a.m.
- Surface Micromachining Module 5 online
- -----
- Today:
- Reading: Senturia Chpt. 3, Jaeger Chpt. 11,  
 Handouts: "Surface Micromachining for  
 Microelectromechanical Systems", "Etch Rates for  
 Micromachining—Part II"
- Lecture Topics:
  - ↳ Polysilicon surface micromachining
  - ↳ Stiction
  - ↳ Residual stress
  - ↳ Topography issues
  - ↳ Nickel metal surface micromachining
  - ↳ 3D "pop-up" MEMS
  - ↳ Foundry MEMS: the "MUMPS" process
  - ↳ The Sandia SUMMIT process
- -----
- Last Time:
- Started into Module 5
- Now continue ...

Microstructure Stiction

Surface Tension

molecule @ liquid surface experiences a net inward force

Liquid Surface

molecule under the liquid surface  
 pulled in all directions  
 ↓  
 net force is zero

attractive forces w/ neighbors

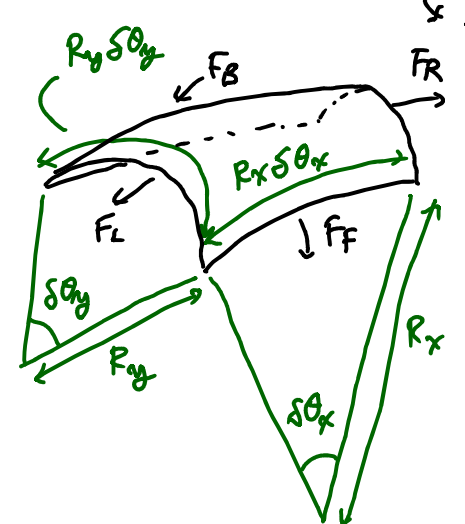
Equilibrium (nothing moves) → forces balanced out by liquid's resistance to compression!

⇒ Result: liquid squeezes to achieve the smallest surface area (smallest energy state)

Surface Curvature & Pressure

No pressure difference  
 ↓  
 surface remains flat

⇒ upon introduction of a differential pressure



surface curves to generate a net normal force to maintain equilibrium against the pressure

Young-Laplace Equation governs the shape of the liquid

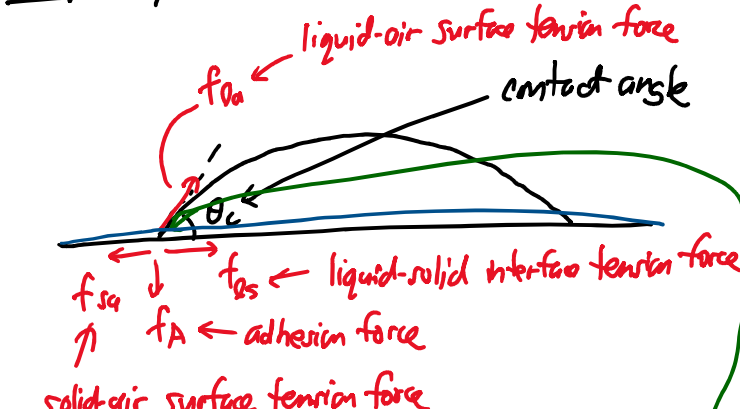
$$\Delta p = \gamma \left( \frac{1}{R_x} + \frac{1}{R_y} \right)$$

where  $\Delta p \hat{=}$  pressure difference  
 $\gamma \hat{=}$  surface tension (force/length)  
 $R_x$  &  $R_y \hat{=}$  radii of curvature

Contact Angle → governed by a balance of surface tensions  
 really a property dependent on the interface between different materials

3

Example. Hydrophilic Droplet



liquid-air surface tension force  $f_{la}$   
 contact angle  $\theta_c$   
 liquid-solid interface tension force  $f_{ls}$   
 solid-air surface tension force  $f_{sa}$   
 adhesion force  $f_A$

Equilibrium: ① horizontal forces cancel } @ the contact pt.  
 ② vertical forces cancel

$$f_A = f_{la} \sin \theta_c$$

$$f_{sa} = f_{ls} + f_{la} \cos \theta_c \rightarrow \gamma_{sa} = \gamma_{ls} + \gamma_{la} \cos \theta_c$$

$f \propto \gamma$

Relationship between surface tensions can be captured by contact angle.

4

Example Two Plates  
 (cross-section)

wetted area = A

Top Plate

Bottom Plate

liquid

$g$

$\frac{g}{2}$

$\theta_c$

$r$

$F$

$F$

Laplace Equation

Pressure Difference @ the Liquid-Air Interface

surface tension @ the liquid-air interface

radius of curvature of the liquid [(-) if convex]

$$\Delta P_{la} = \frac{\gamma_{la}}{r}$$

$r = \frac{-g/2}{\cos \theta_c} \Rightarrow F = -\Delta P_{la} A = \frac{2A \gamma_{la} \cos \theta_c}{g}$

Force needed to keep the plates apart.

$\Rightarrow$  (+) force means (-) Laplace pressure

5

$F_{spring} = k \Delta g$  stiffness

$\Delta g$

$g$

$A = a \cdot b$

spring

Remarks.

① To prevent stiction:

- $\rightarrow$  reduce A (wetted area)
- $\Rightarrow$  reduce  $\gamma_{la} \rightarrow$  choose the right liquids
- $\Rightarrow$  make  $g =$  large
- $\Rightarrow$  increase  $k \rightarrow$  make things thicker
- $\Rightarrow \theta_c > 90^\circ$  (choose the right solid/liquid combination)

6

Liquid	Solid	Contact angle
water	soda-lime glass	0°
ethanol	lead glass	
diethyl ether	fused quartz	
carbon tetrachloride		
glycerol		
acetic acid		
water	paraffin wax	107°
	silver	90°
methyl iodide	soda-lime glass	29°
	lead glass	30°
	fused quartz	33°
mercury	soda-lime glass	140°
Some liquid-solid contact angles <sup>[5]</sup>		